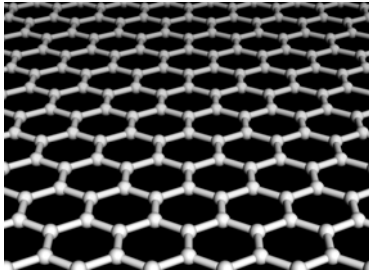


## Part II-2

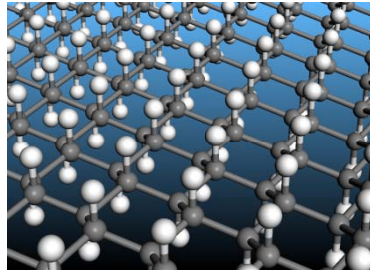
# **Bilayer graphene and Gap Opening**

# Gap opening

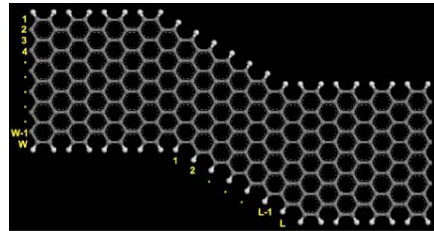
Graphene ( $sp^2$ )



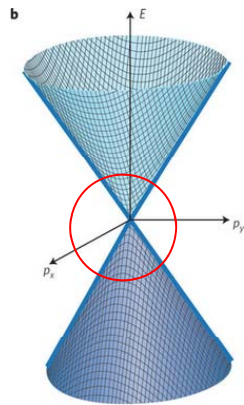
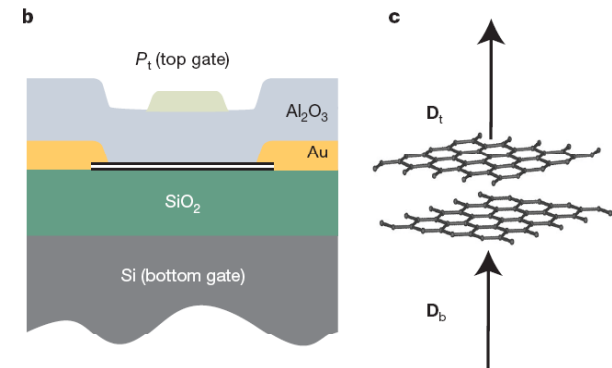
Graphane( $sp^3$ )



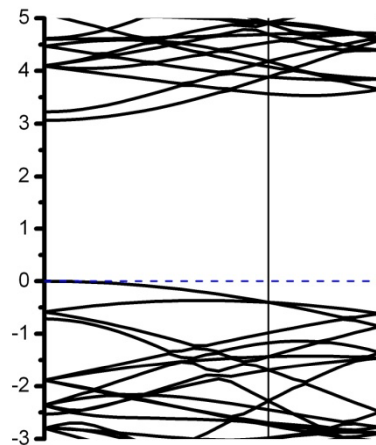
Graphene ribbon



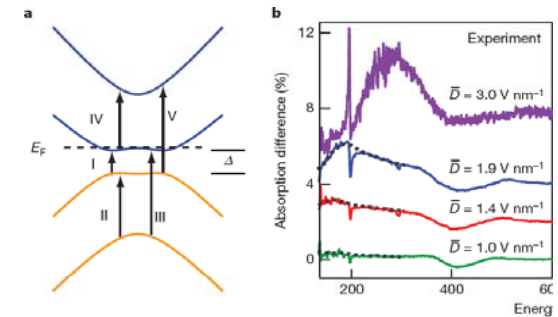
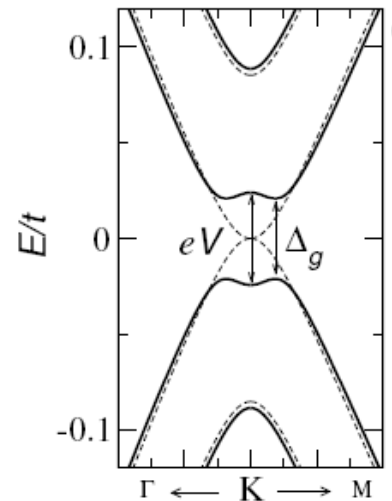
Under applied E field



No band gap



Wide band gap

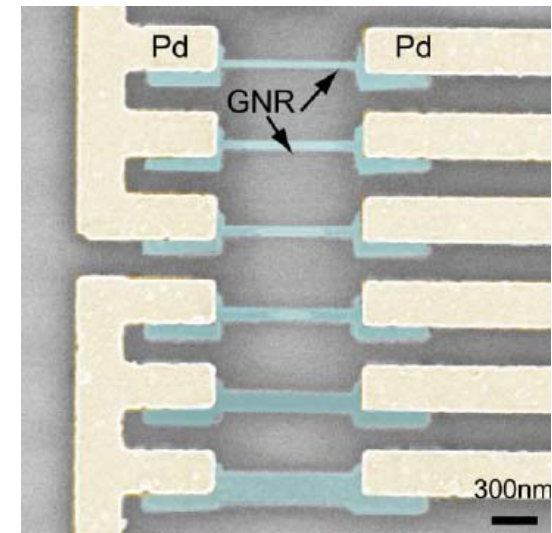
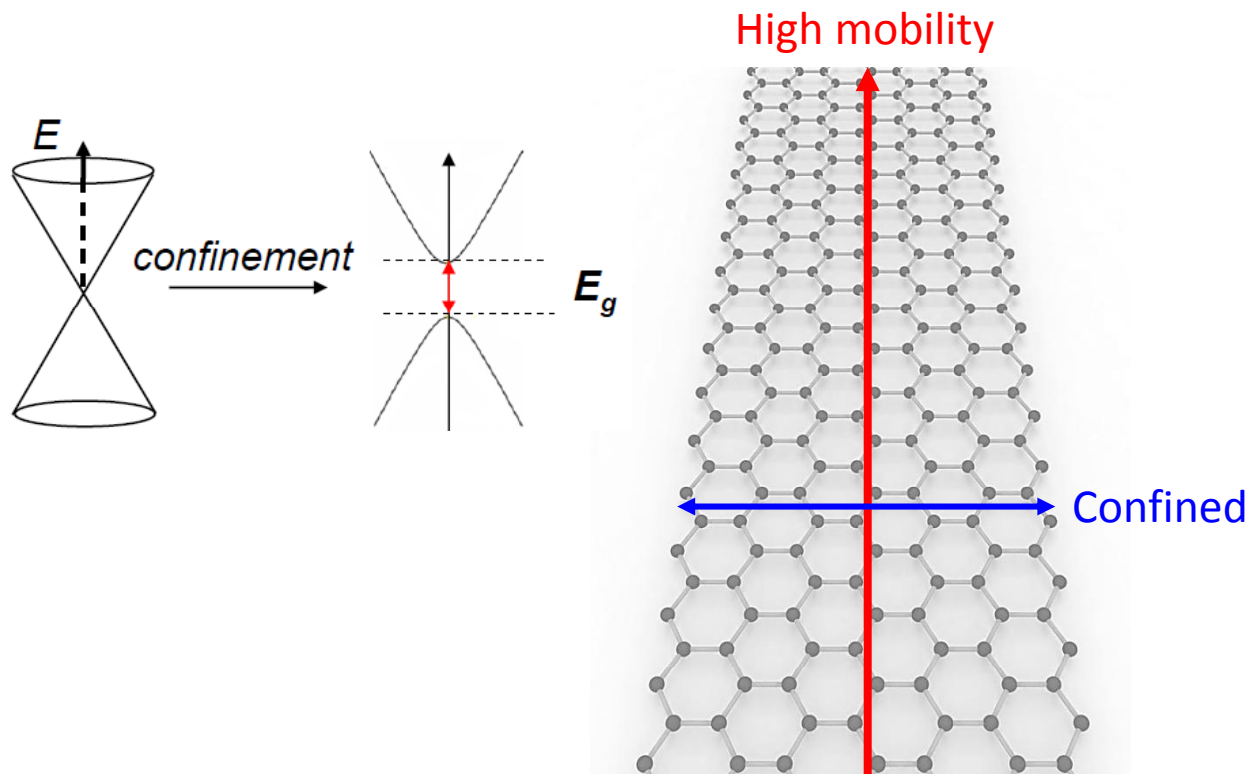


Nature, 2009, Vol. 459, p820

Elias et al,  
Science, 323, 5914, 2009

# Electronic Structures of GNRs

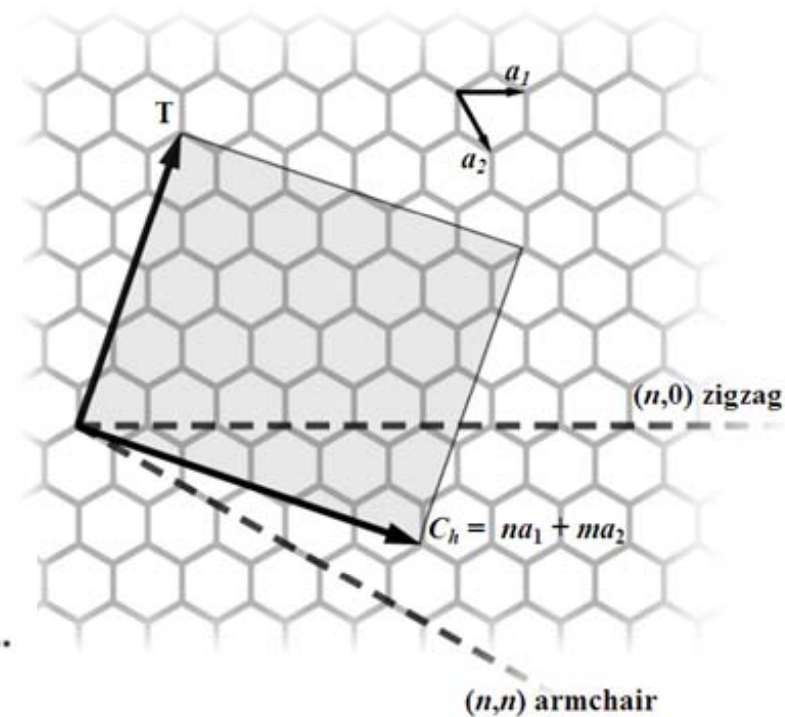
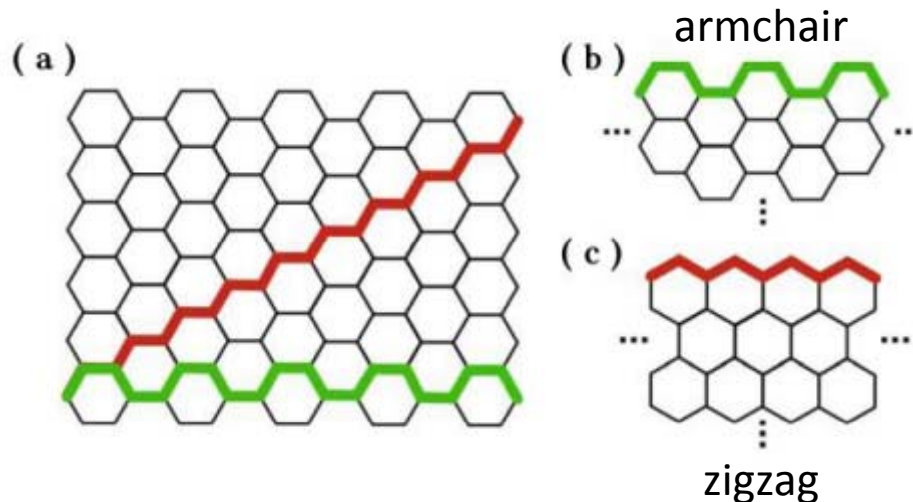
- Open a gap in graphene-graphene nanoribbon (GNR)
- High mobility along the specific direction
- Quantum confinement effect



*Physical E*, 40, 228, (2007)

# Graphene Nanoribbon (GNR)

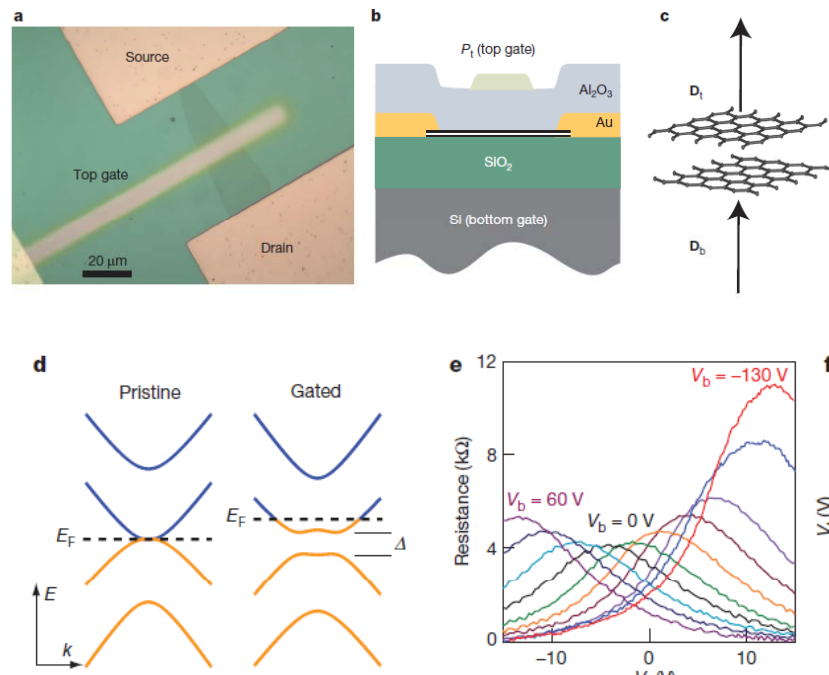
- Structure
  - Graphene strips
  - Chirality
    - Two types of edge



Some of armchair GNRs are semiconducting  
metallic:  $n = 3m+2$   
semiconducting:  $n = 3m$  or  $3m+1$

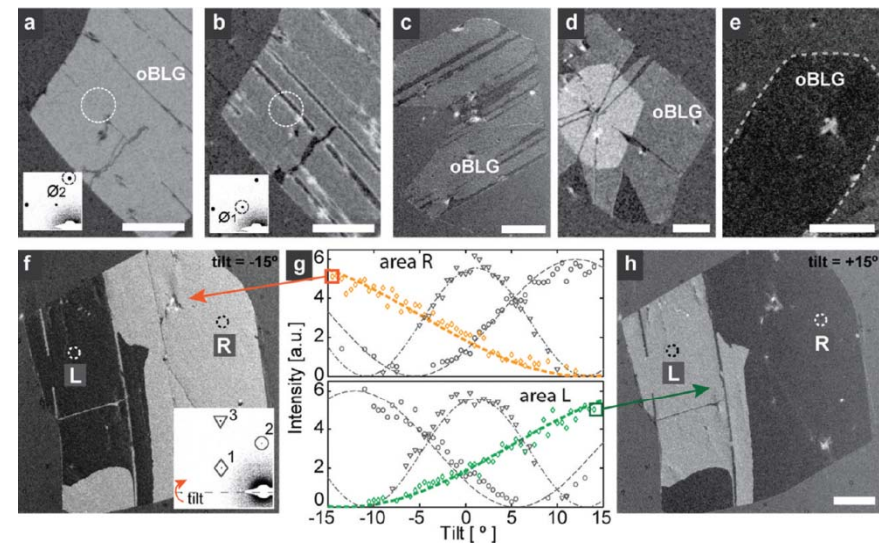
# Gap-opening of Bilayered-graphene

Mechanically exfoliated garpehene



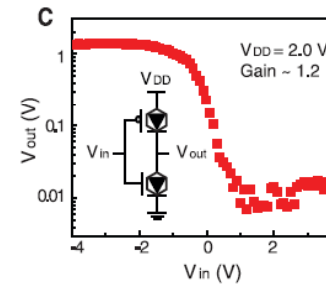
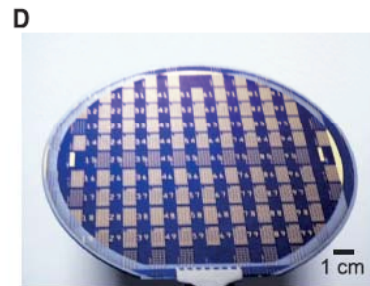
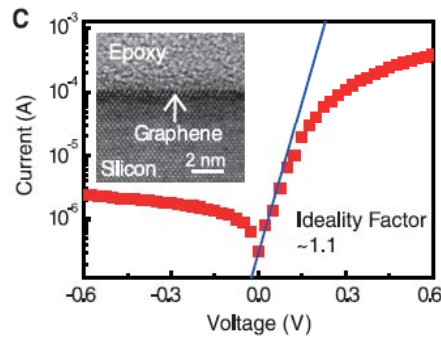
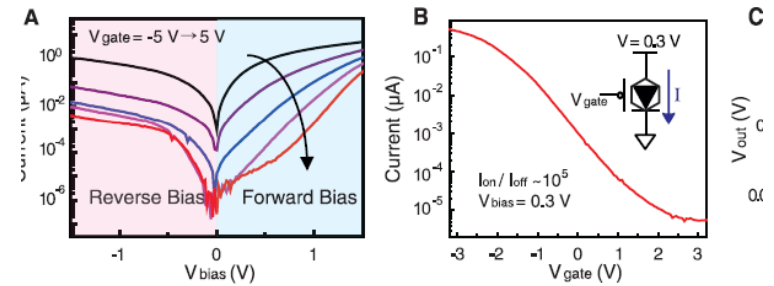
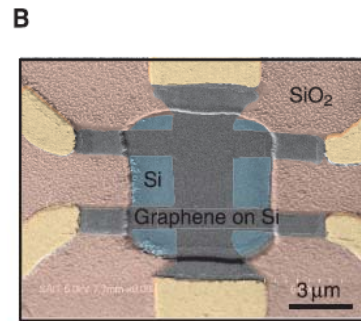
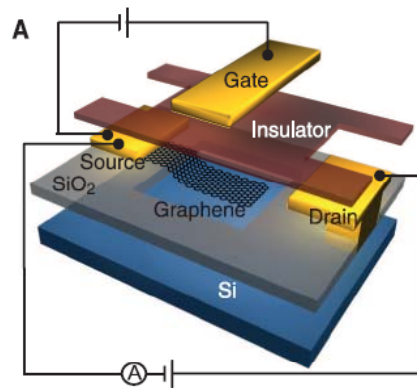
Nature, 2009, Vol.459, p820

CVD-graphene



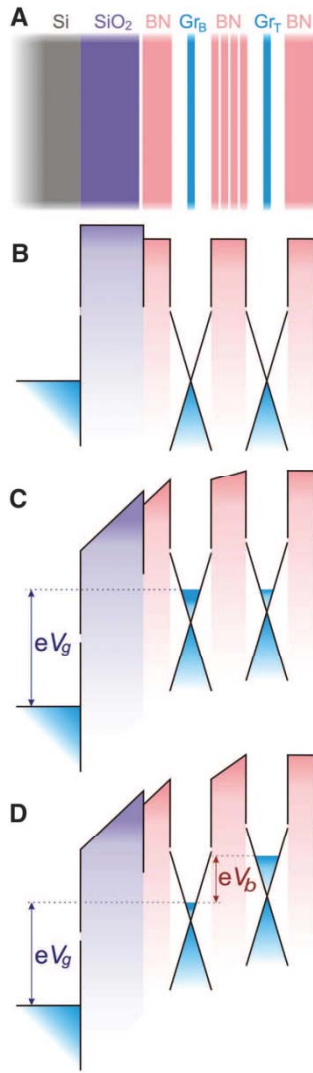
Nano Letters, 2012, 12, 1609

# Graphene Barristor- a gate controlled Schottky barrier

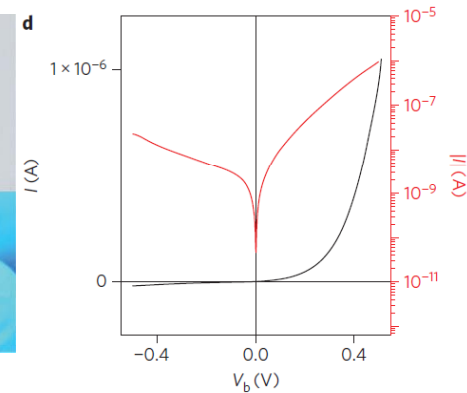
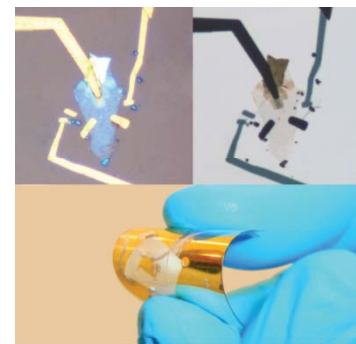
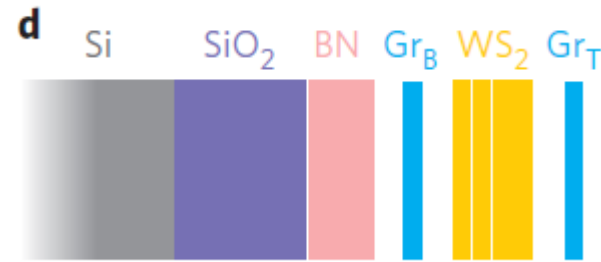
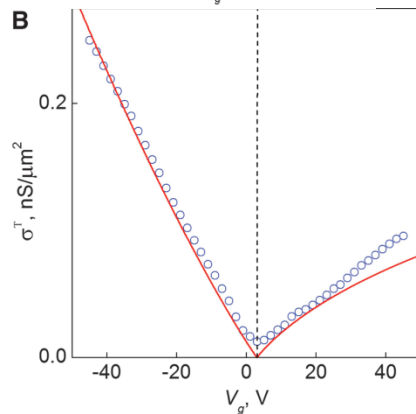
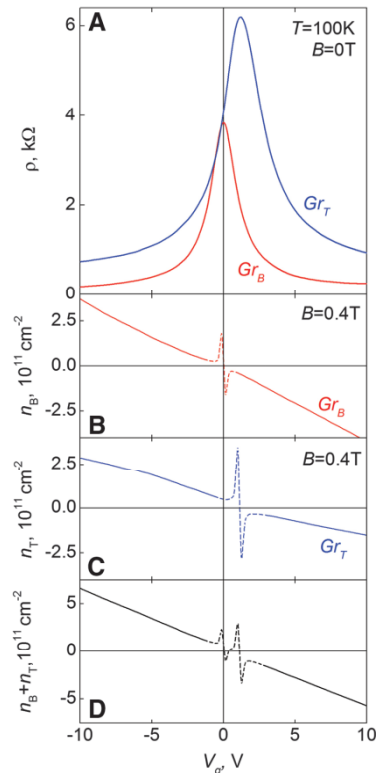


Adjusting graphene workfunction  
High on/off ratio  $\sim 10^5$

# Vertical transnsitor



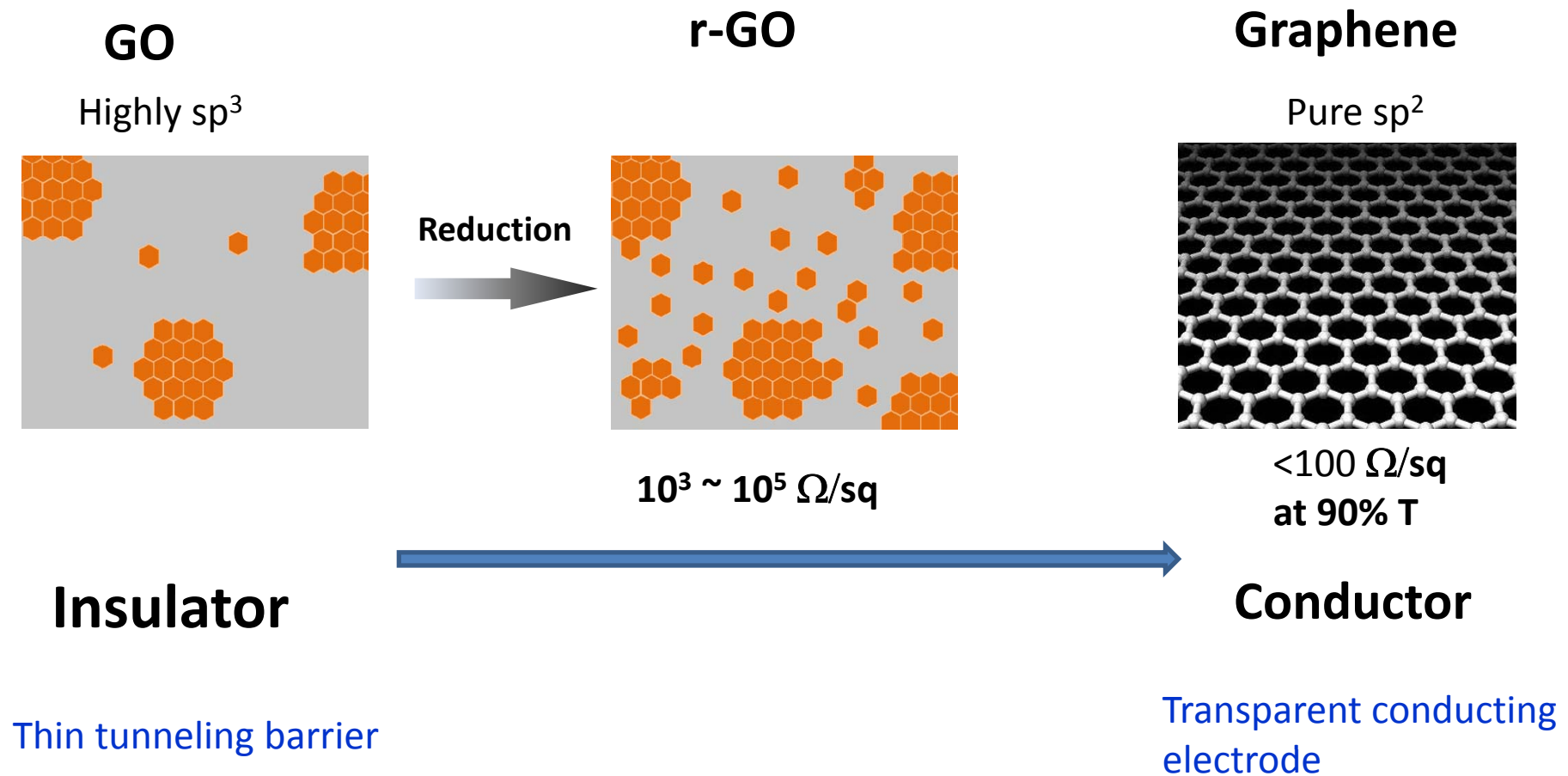
*Science*, 335 947-950 (2012)



*Nature Nanotechnol.* 8, 100–103 (2013)

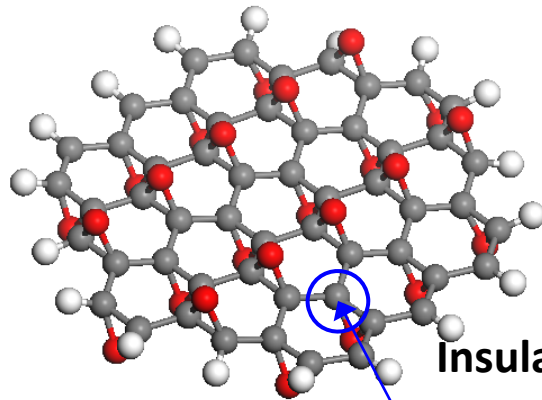
- Combine kinds of 2-D material
- Control tunneling barrier through ultra-thin barrier
- Very Difficult to fabricate

# Tunable conduction in graphene based materials



# Atomic and electronic structure of GO/Graphene

GO nanosheet

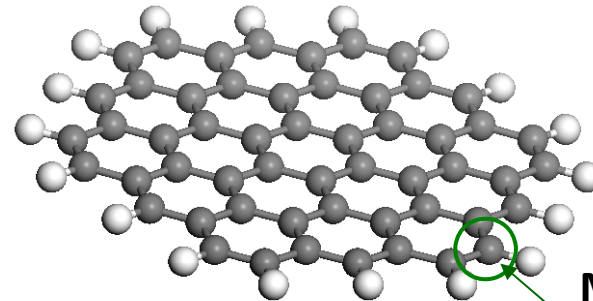


Ideal structure



Reduction

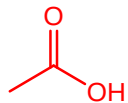
Graphene



Metallic-like  
 $sp^2$  (planar)

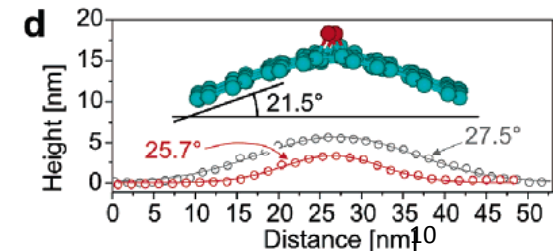
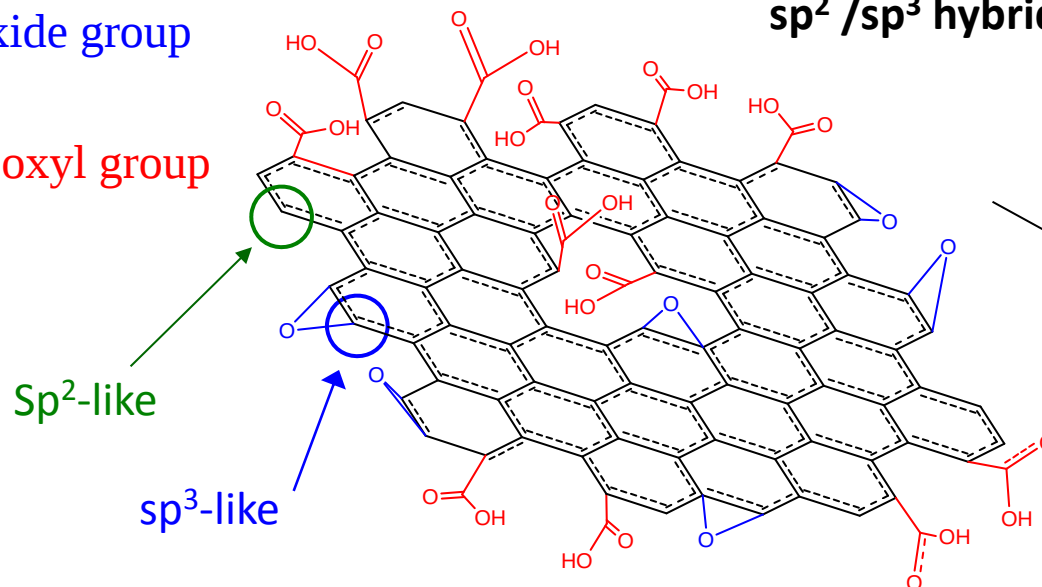


Epoxide group



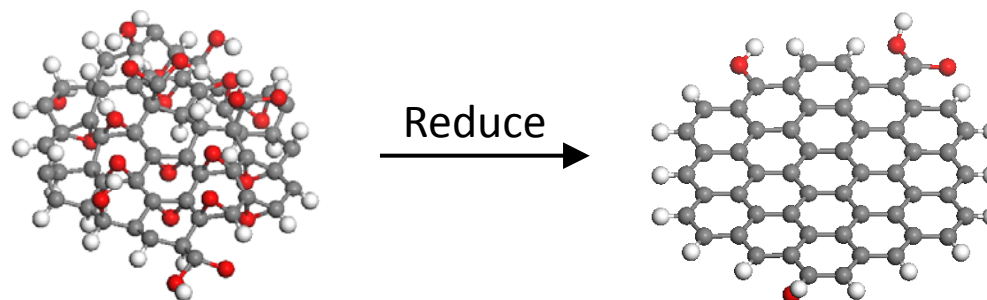
Carboxyl group

$sp^2$  /  $sp^3$  hybrid ( Real structure)

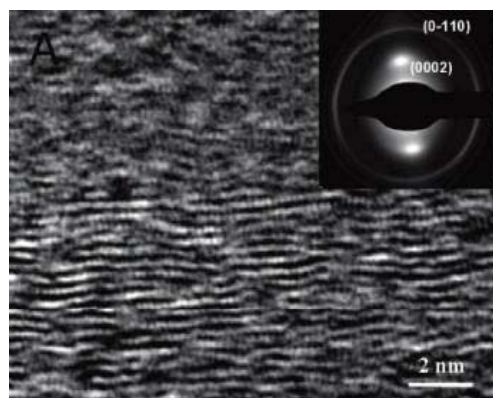


Akasay et al, J of Phys. Chem. B Lett., 110, 8535, 2006

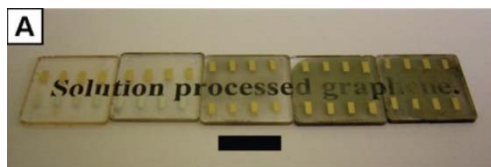
# Preparation of reduced GO



## UHV thermal anneal

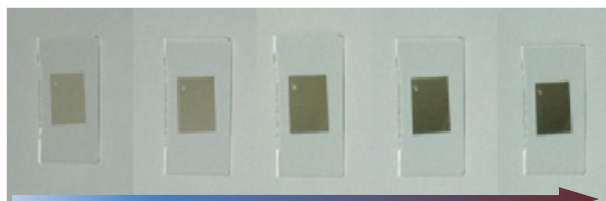
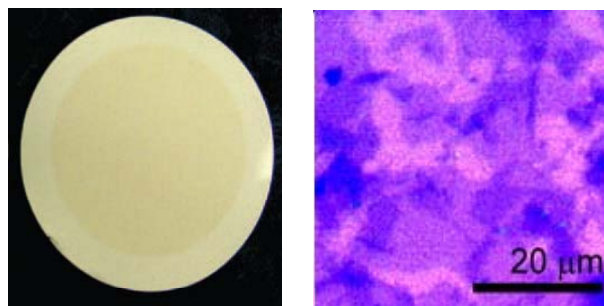


X. Wang *et al.* *Nano Lett.* 8 323-327 (2008)



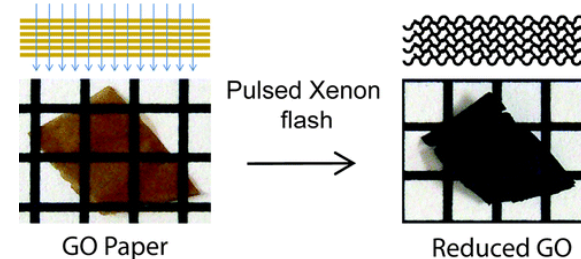
H. A. Becerril *et al.* *ACS Nano* 2 463-370 (2008)

## $N_2H_4$ vapor



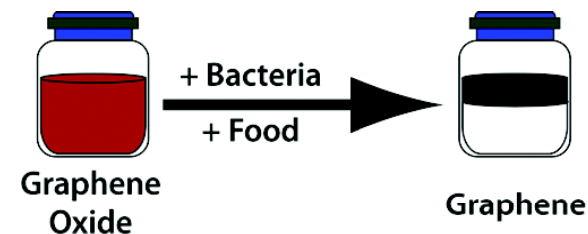
Eda *et al.* *Nature Nanotech.* 3 270-274 (2008)

## Flash light reduction



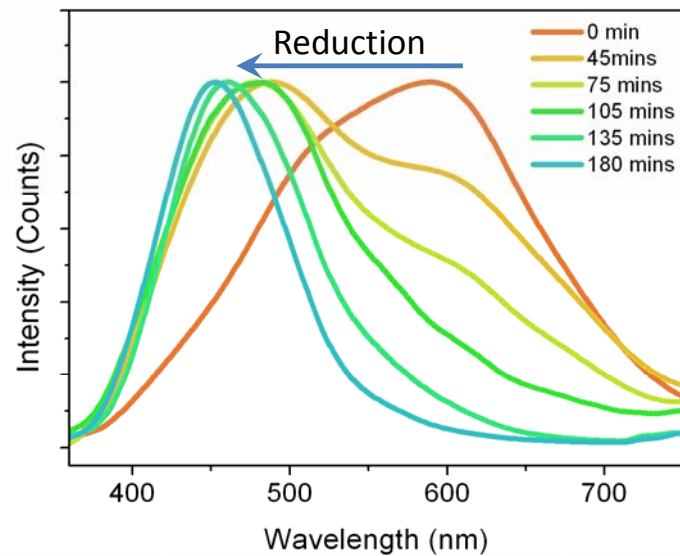
L. J. Cote *et al.* *JACS* 131 11027-11032 (2009)

## Biological reduction

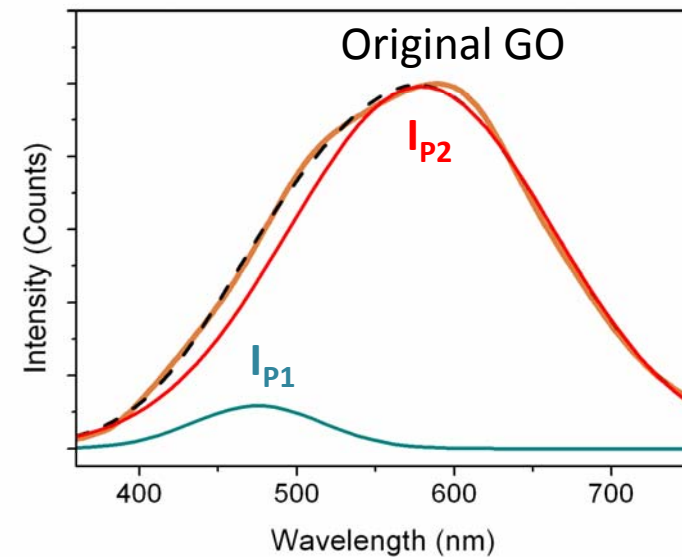
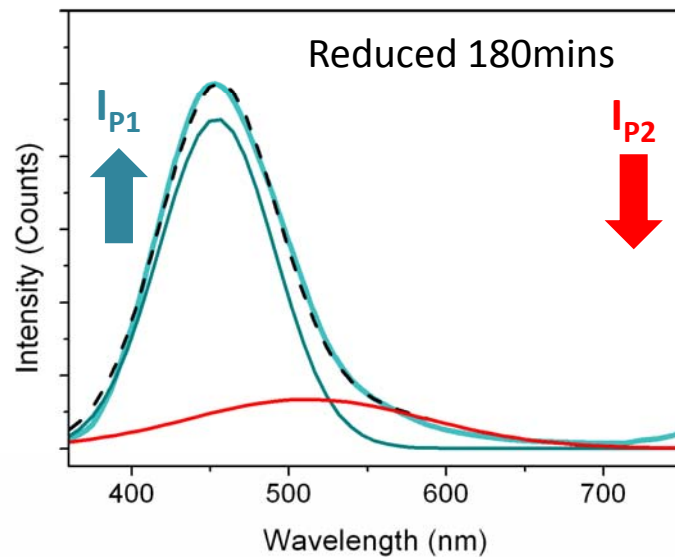


E. C. Salas *et al.* *ACS Nano* 4 4852-4856 (2010)

# Evolution of photoluminescence of graphene oxide with reduction (I)

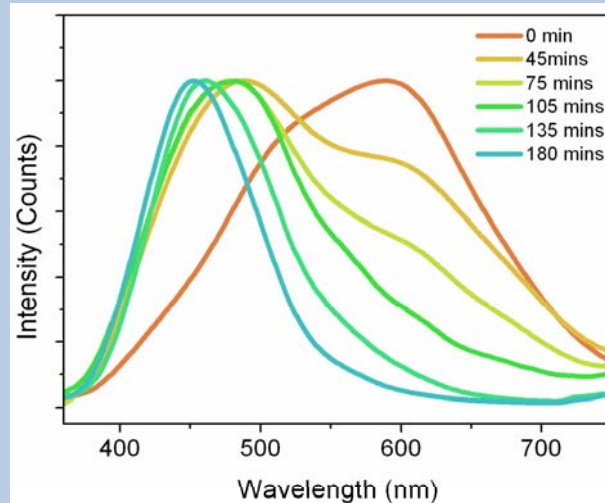
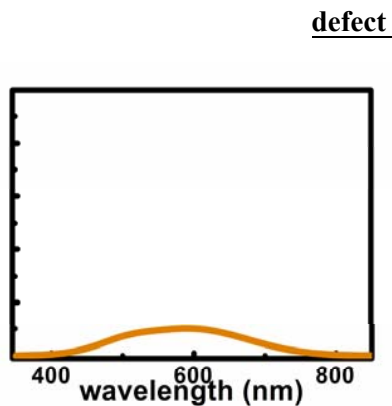


*Angewandte Chemie Int. Ed.*, 51, 6662, (2012)

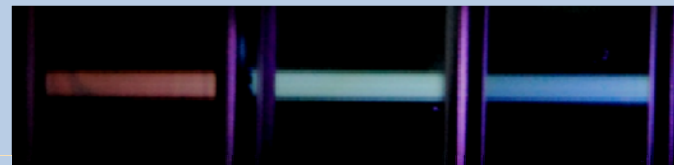


# Evolution of photoluminescence of GO and r-GO

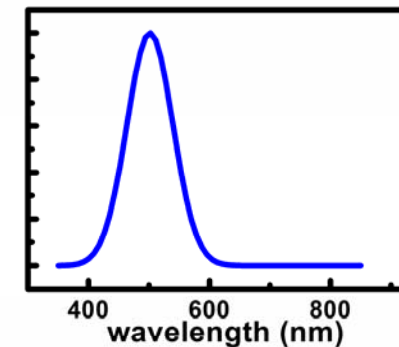
*Angewandte Chemie Int. Ed.*,  
51, 6662, (2012)



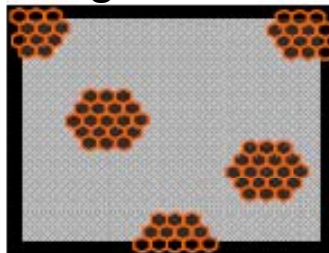
Cluster-like state  
small sp<sup>2</sup> clusters or  
atomic chains)



**Tunable PL**



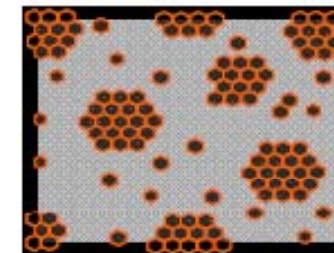
**Original GO**



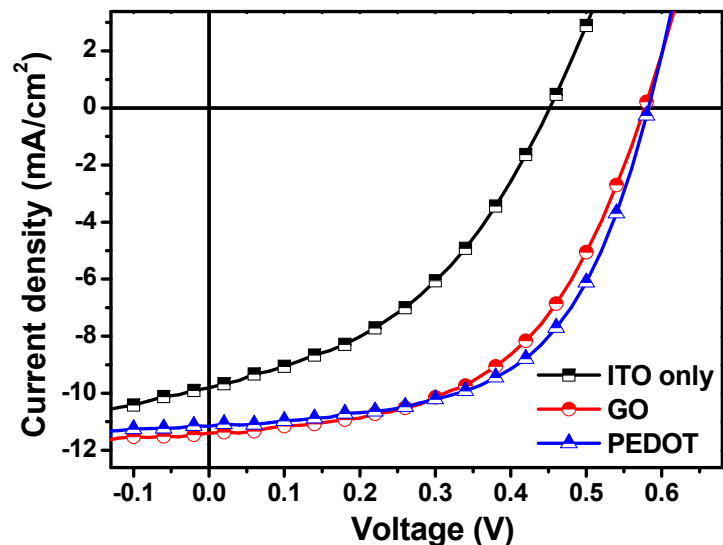
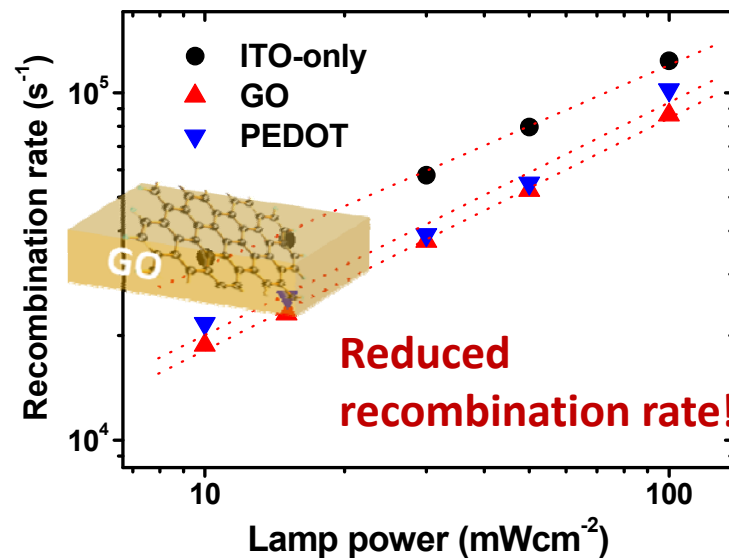
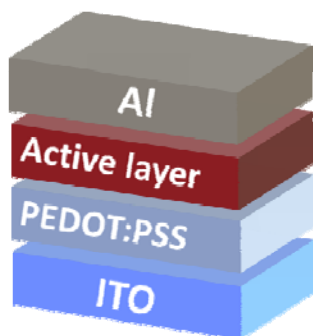
**Reduction**

**Conductivity**

**Reduced GO**



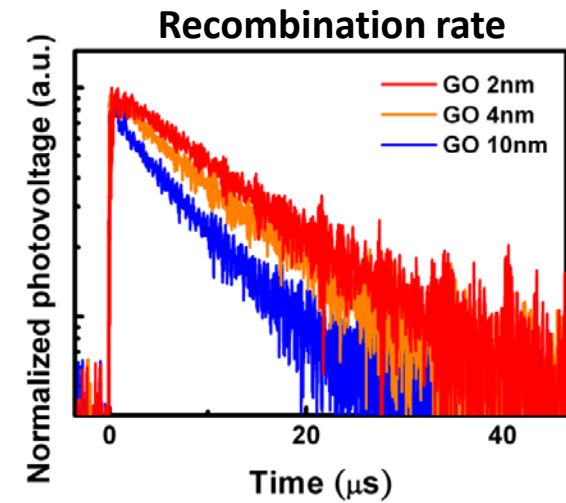
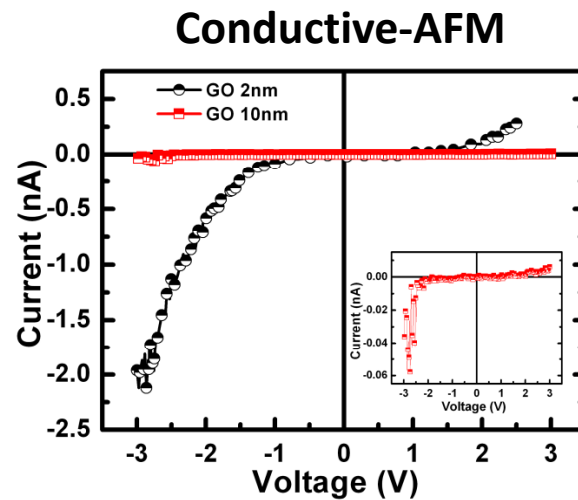
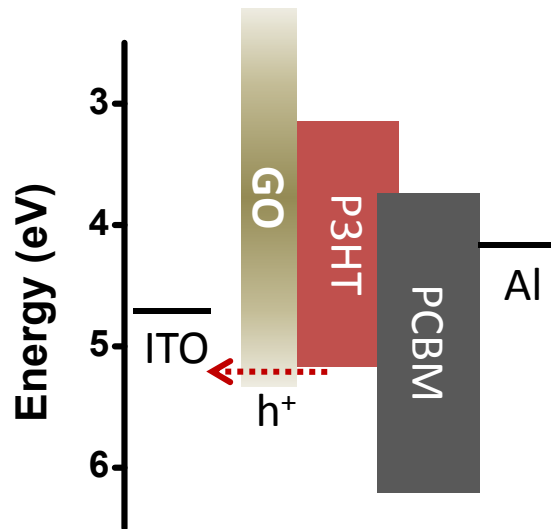
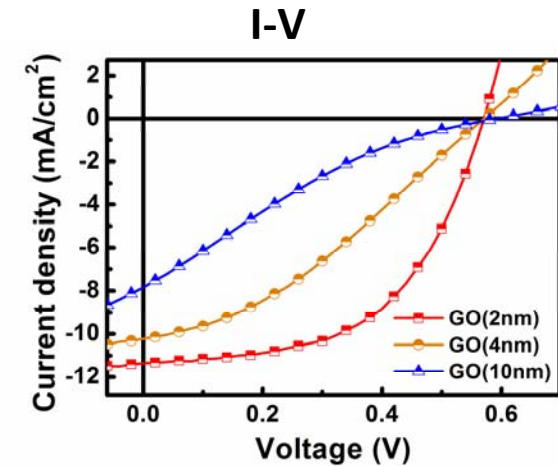
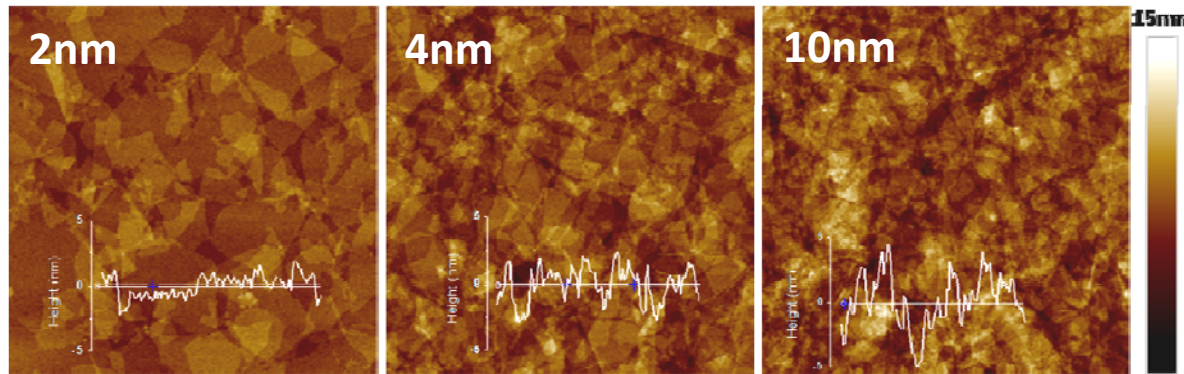
# GO as a hole transport layer (replace PEDOT:PSS)



|           | Voc (V) | Jsc (mAcm <sup>-2</sup> ) | FF (%) | η (%) |
|-----------|---------|---------------------------|--------|-------|
| ITO only  | 0.45    | 9.84                      | 41.51  | 1.84  |
| ITO/GO    | 0.58    | 11.40                     | 54.34  | 3.53  |
| ITO/PEDOT | 0.58    | 11.15                     | 56.86  | 3.69  |

*ACS Nano, Vol. 4, 3169, (2010)*

# GO act as a thin tunneling barrier for hole transport

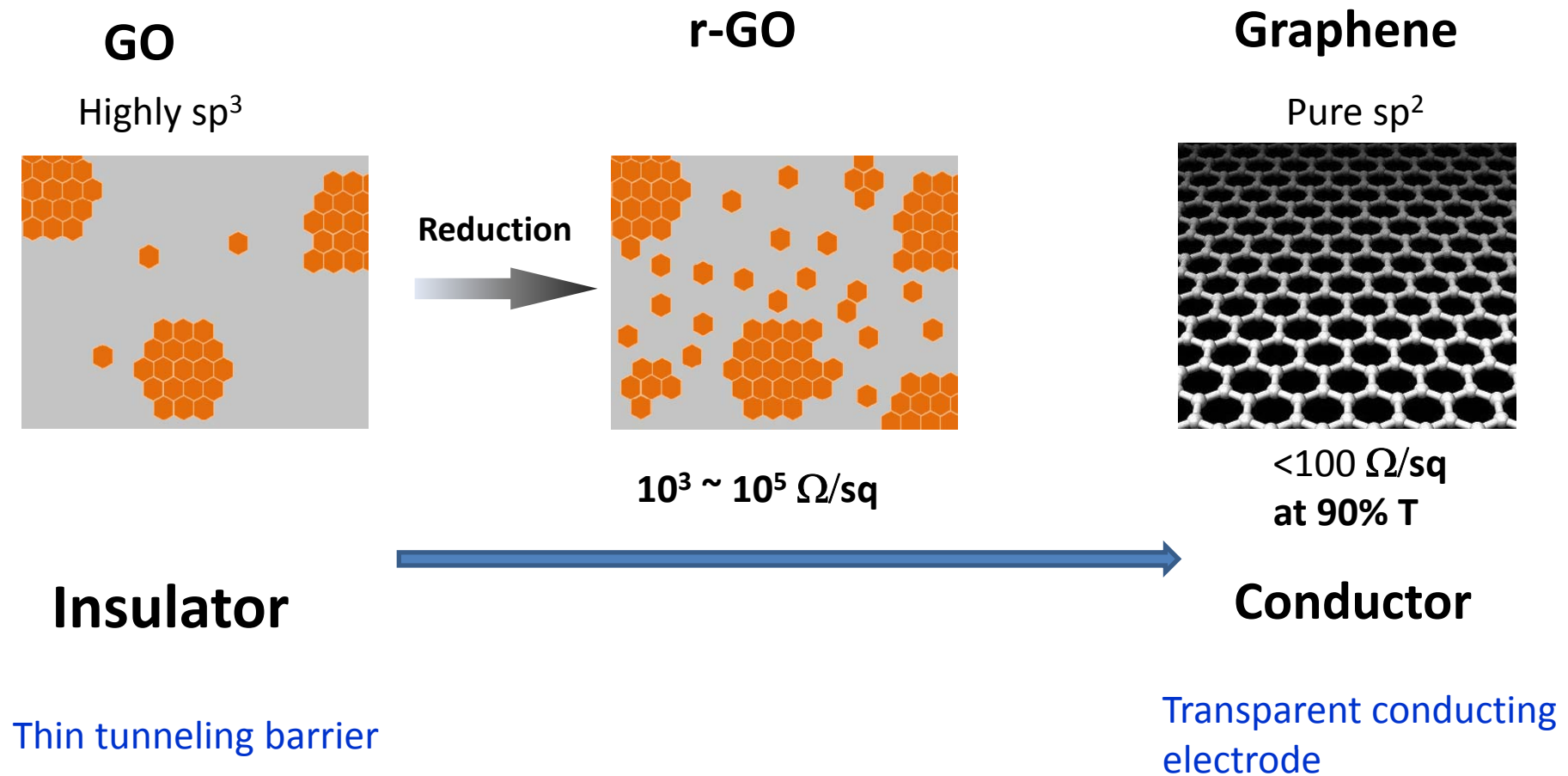


Tunneling current

$$I_{2nm} = 2 \text{ nA}$$

$$I_{10nm} = 0.06 \text{ nA}$$

# Tunable conduction in graphene based materials



Graphene for transparent electrode

# Next generation production ?



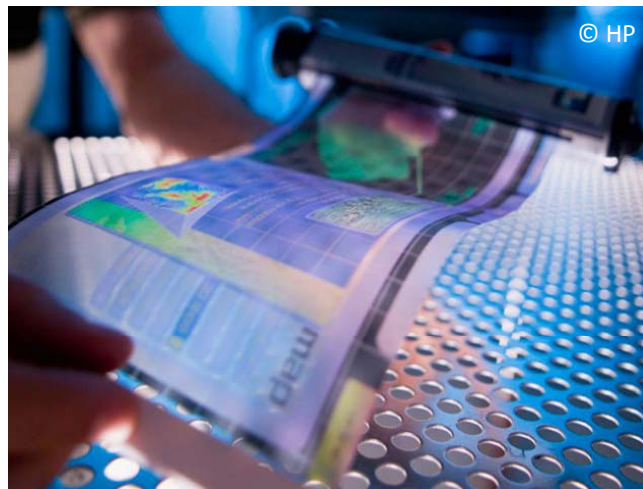
© NOKIA



Household appliances

E-paper

- High conductivity
- High transparency
- Good mechanical property
- Flexible
- Good thermal conductivity
- etc.



© HP



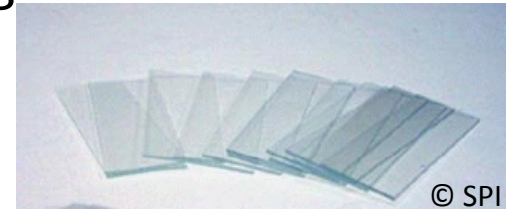
© SAMSUNG

Communication  
production

# Conventional Transparent Electrode

- **ITO (Indium Tin Oxide)**

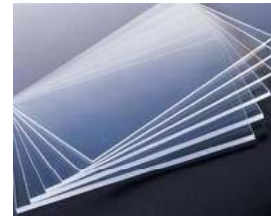
- Scarcity of indium and high manufacturing cost
- Vacuum process
- Relatively brittle



- **Alternative :**

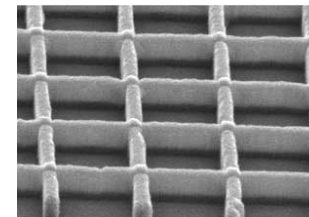
- Conducting oxides

- Al doped ZnO

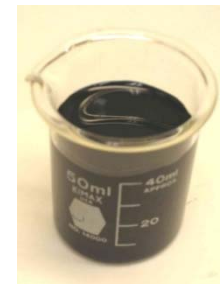
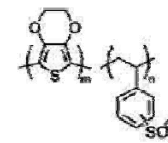


- Metal wire mesh

- Ag nanowire, Cu grid

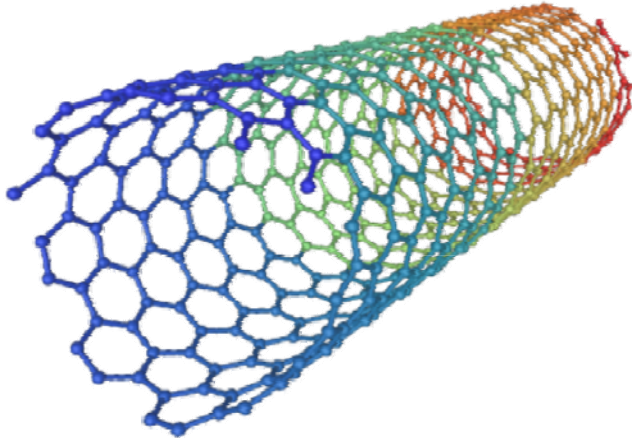


- Conductive polymer, PEDOT:PSS



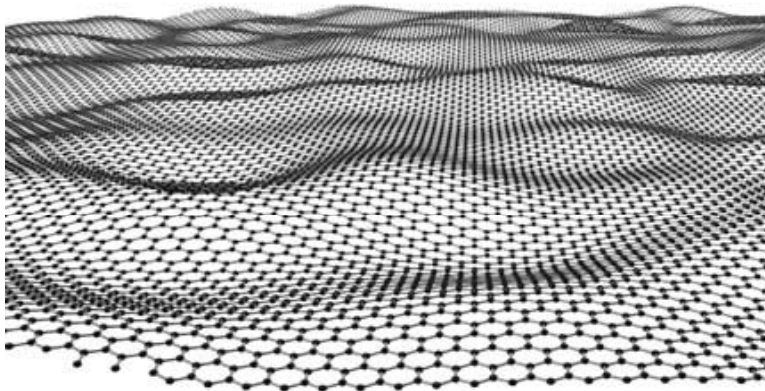
# Low Dimensional Nanocarbon Materials

## Single walled carbon nanotube (1D)



- 1-D material
- Metallic and semiconducting
- High carrier mobility ( $120,000 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ )
- High current carrying capacity
- Excellent mechanical strength

## Graphene (2D)



- $\text{sp}^2$  bonded carbon
- 2-D honeycomb crystal lattice
- High carrier mobility ( $200,000 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ )
- Low electrical resistivity ( $10^{-6} \Omega\cdot\text{cm}$ )  
( $< \text{silver}$ )
- Good mechanical properties

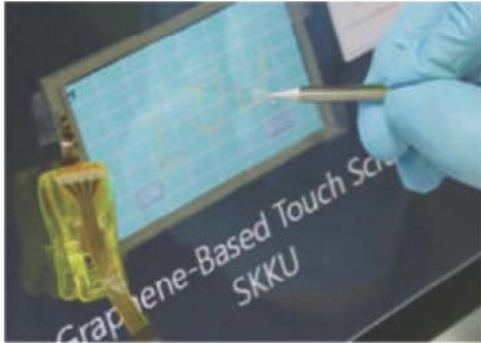
# List of various end applications, their key features, and the suitability of each material

| Application     | Key Features      | Nanotube | Graphene | Metal Nanowires |
|-----------------|-------------------|----------|----------|-----------------|
| Touch Panels    | Flexibility       | o        | o        | x               |
|                 | Patterning        | o        | o        | o               |
|                 | Sheet resistance  | o        | -        | o               |
|                 | Transparency      | -        | -        | o               |
| LCD             | Surface roughness | -        | o        | x               |
|                 | Ionic impurities  | x        | x        | x               |
|                 | Conformal Coating | o        | o        | x               |
|                 | Color/Haze        | o        | o        | x               |
| OLED/Solar Cell | Work Function     | o        | o        | x               |
|                 | Sheet resistance  | x        | x        | o               |
|                 | Surface Roughness | -        | o        | x               |
|                 | Stability         | o        | o        | x               |

("o" = superior, "-" = good, "x" = poor).

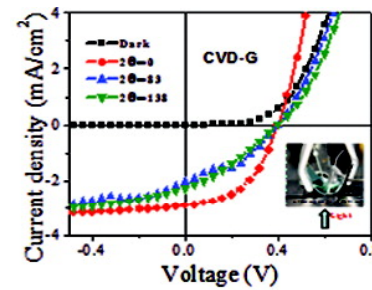
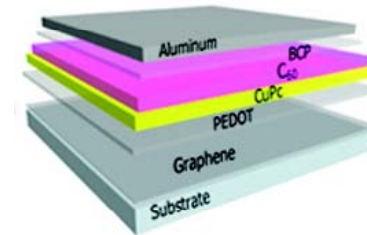
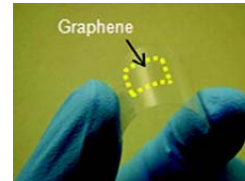
- Transparent conducting electrode

### Touch Panel



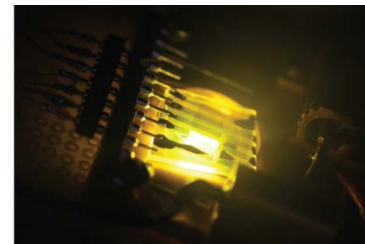
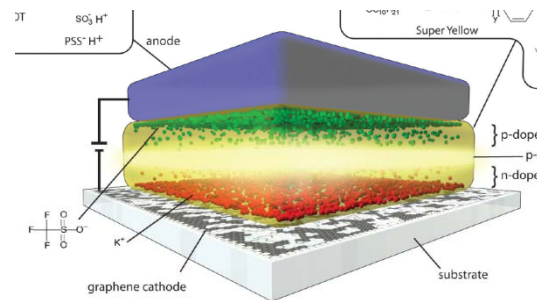
Bae, S. *et al.*  
*Nature Nanotech.* **4**, 574–578 (2010).

### Solar Cell



De Arco, L. G. *et al*  
*ACS Nano* **4**, 2865–2873 (2010)

### Light Emitting Diode



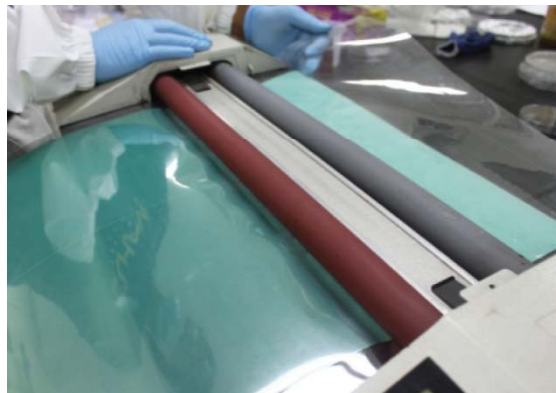
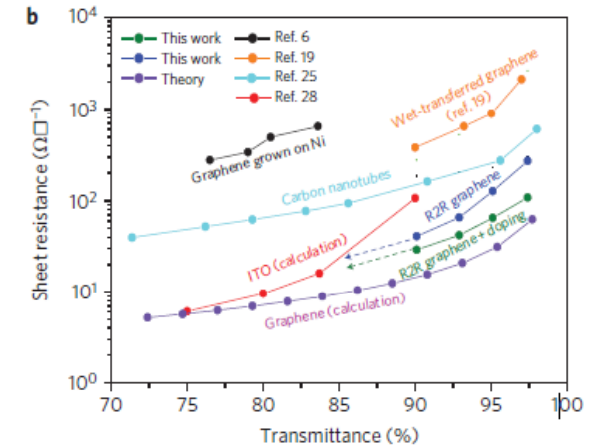
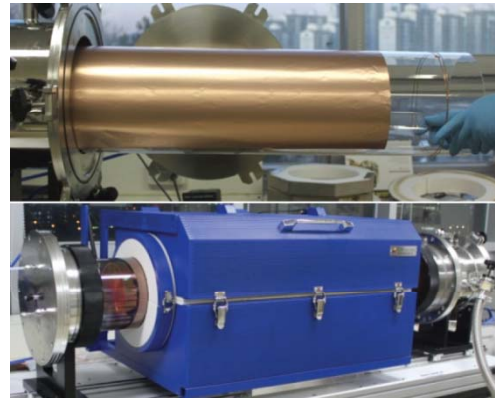
Matyba, P. *et al.*  
*ACS Nano* **4**, 637–642 (2010).

# Large area CVD grown graphene for optoelectronic applications

nature nanotechnology LETTERS  
PUBLISHED ONLINE: 20 JUNE 2010 | DOI: 10.1038/NNANO.2010.132

## Roll-to-roll production of 30-inch graphene films for transparent electrodes

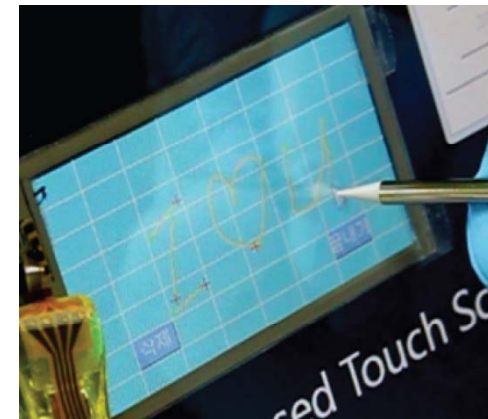
Jong-Hyun Ahn and Byung Hee Hong *et al.*\*



Roller printing

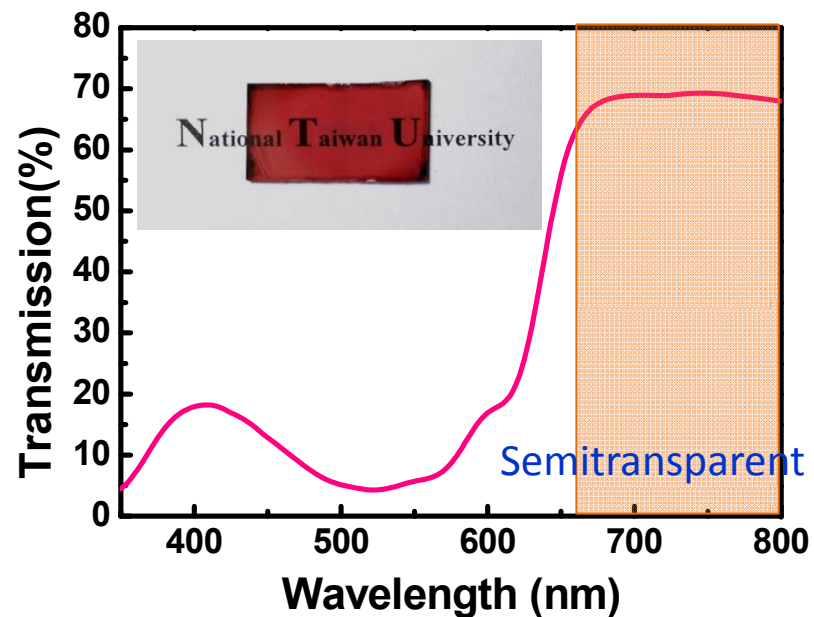
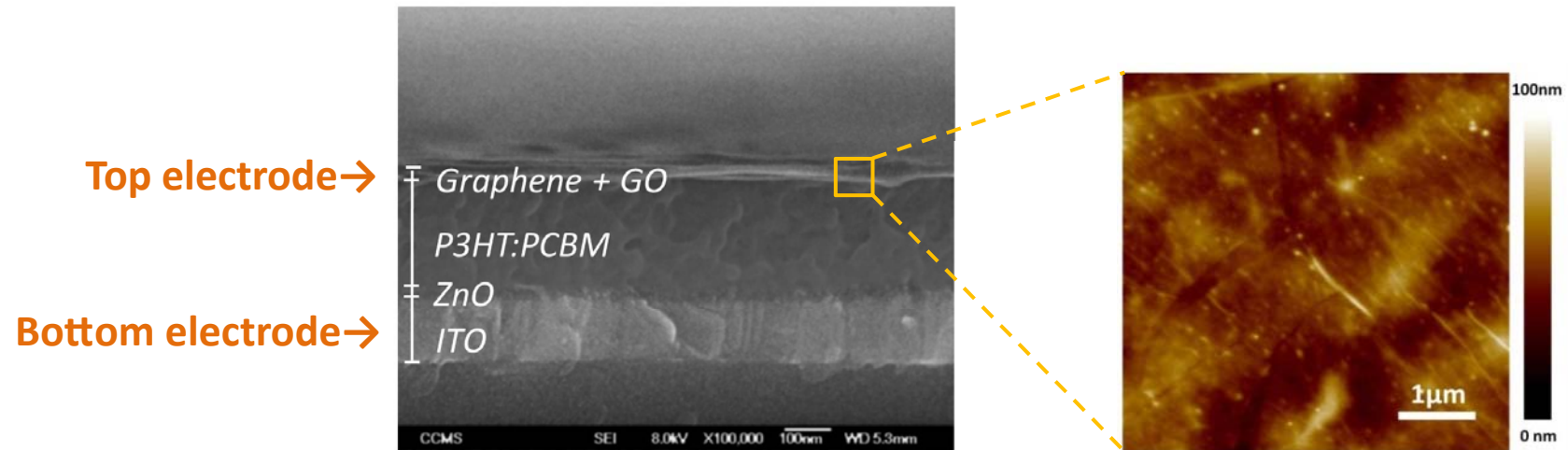


Scale up



Touch panel

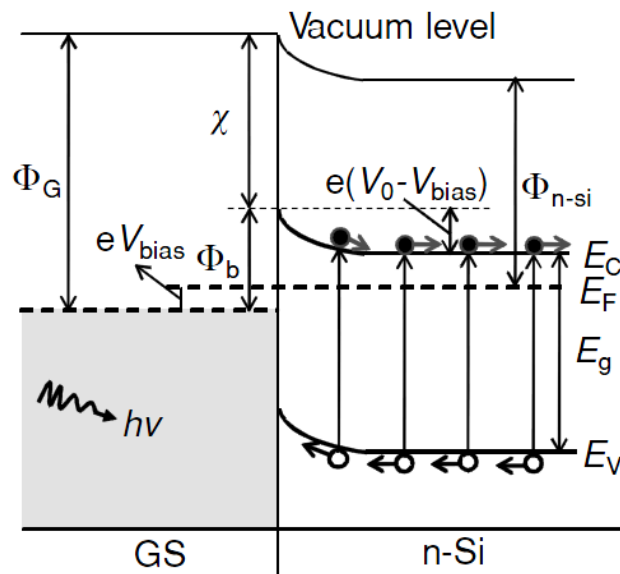
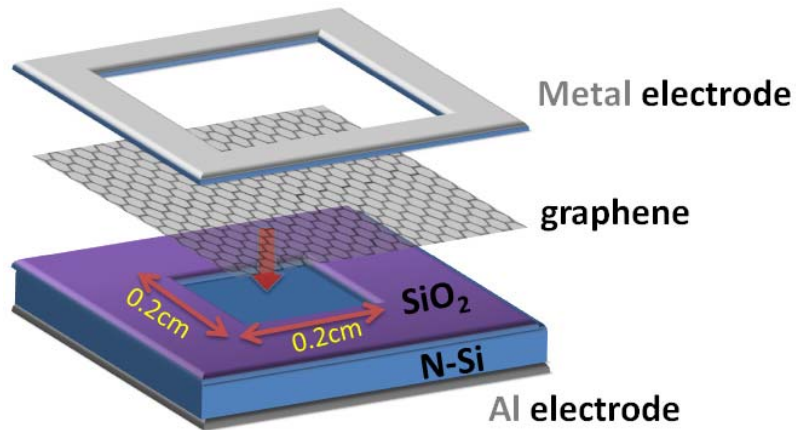
# Top laminated graphene electrode in a semitransparent polymer solar cell



*ACS Nano ,Vol.5, 6564, (2011)*

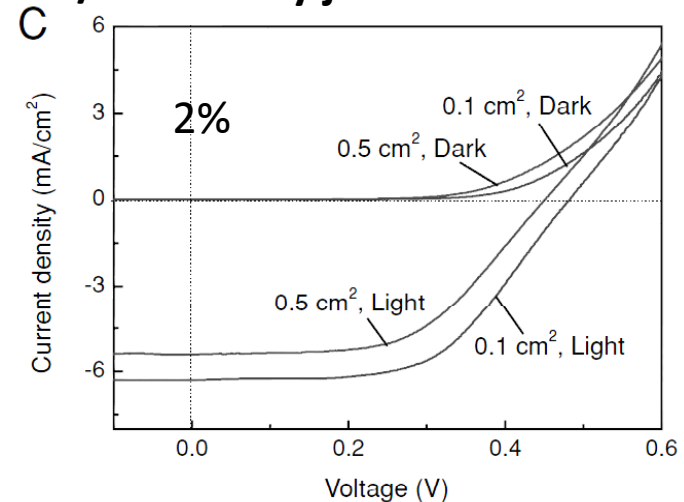
# Si/Graphene junction solar cell

## Device structure

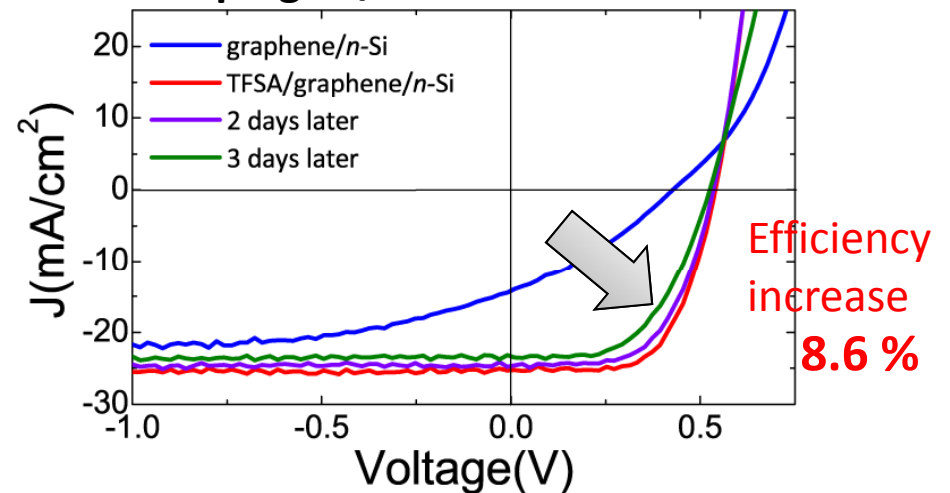


Energy diagram of the forward-biased GS/n-Si Schottky junction

## Gr/Si schottky junction solar cell



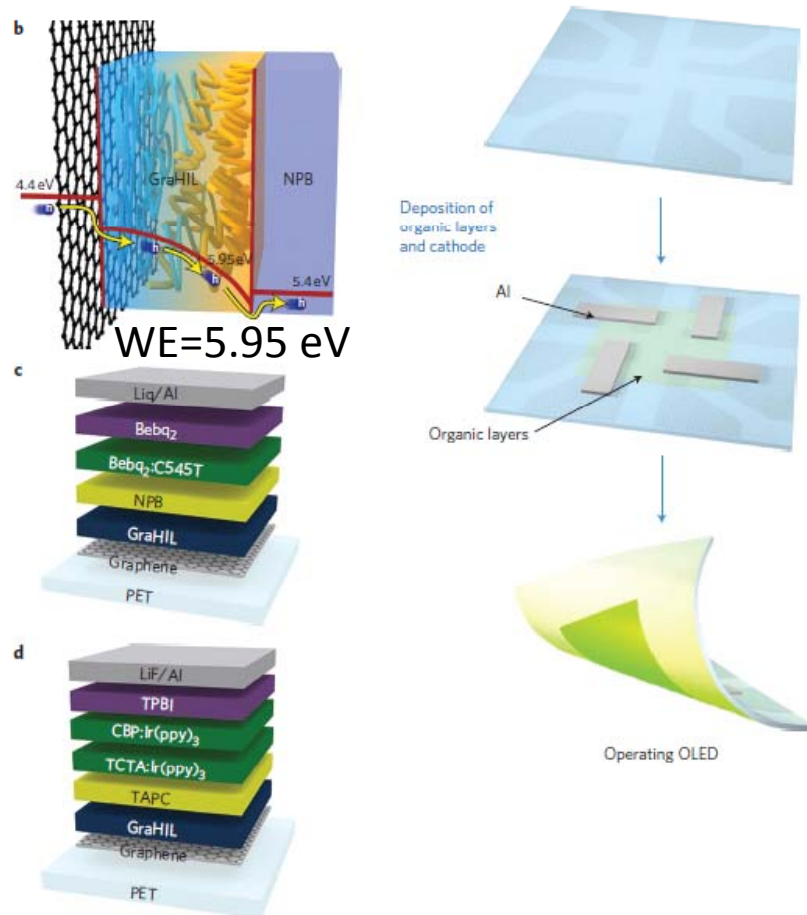
## P doping Gr/Si



Li. X. et. al., *Adv. Mater.* **2010**, 22, 2743–2748  
Miao, X. Et. al., *Nano Lett.* **2012**, 12, 2745–2750

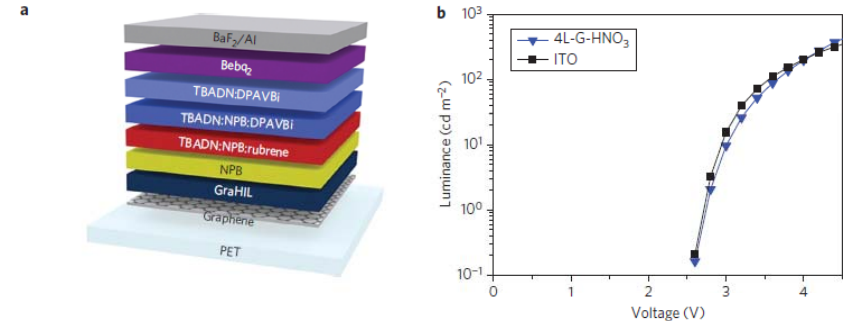
# The Future development of large area CVD graphene in OLED applications

4-layered  $\text{HNO}_3$  doped Gr  $\sim 30 \Omega/\square$

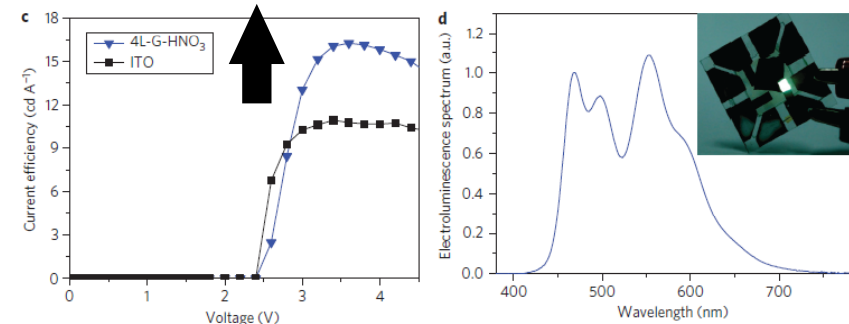


**Gra HIL:** Self-organized gradient hole injection layer

T.H. Han, et al., *Nature Photonics*, **6**, 105–110, (2012).

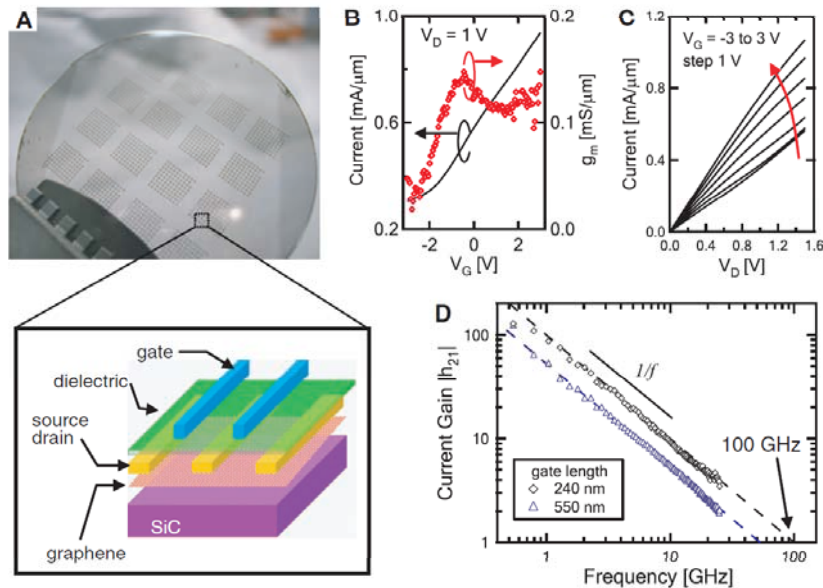


The white OLEDs with the graphene anode exhibited a much higher the ITO anode



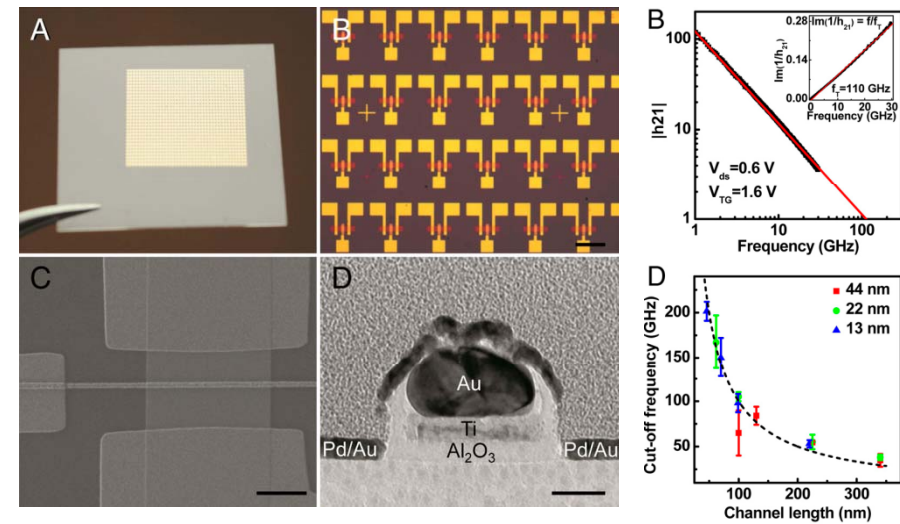
# **Graphene for other applications**

# Graphene for Radio Frequency (RF) Transistor



*Science*, 327 , 662 (2010)

up to **100** GHz (cut-off frequency)



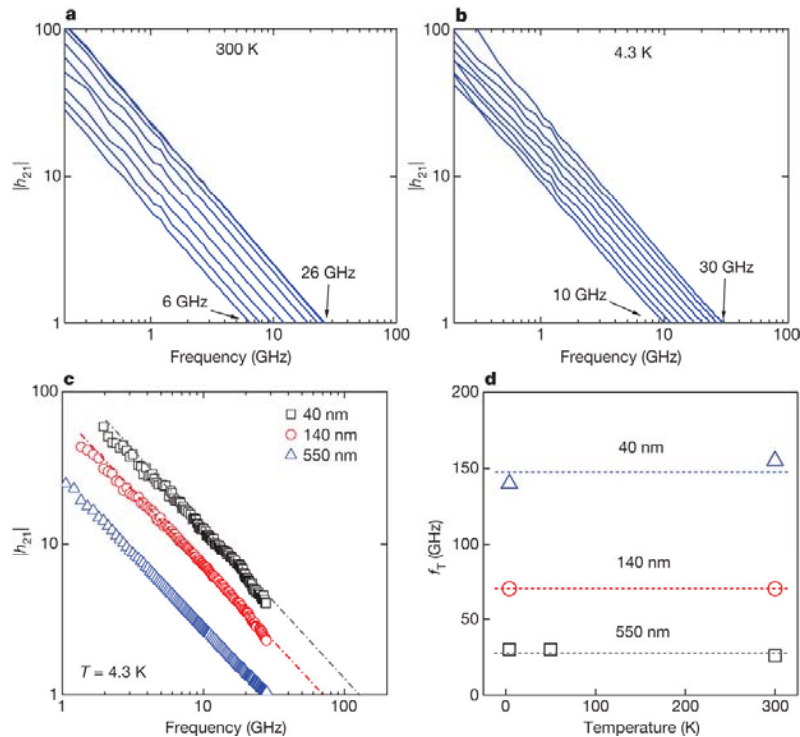
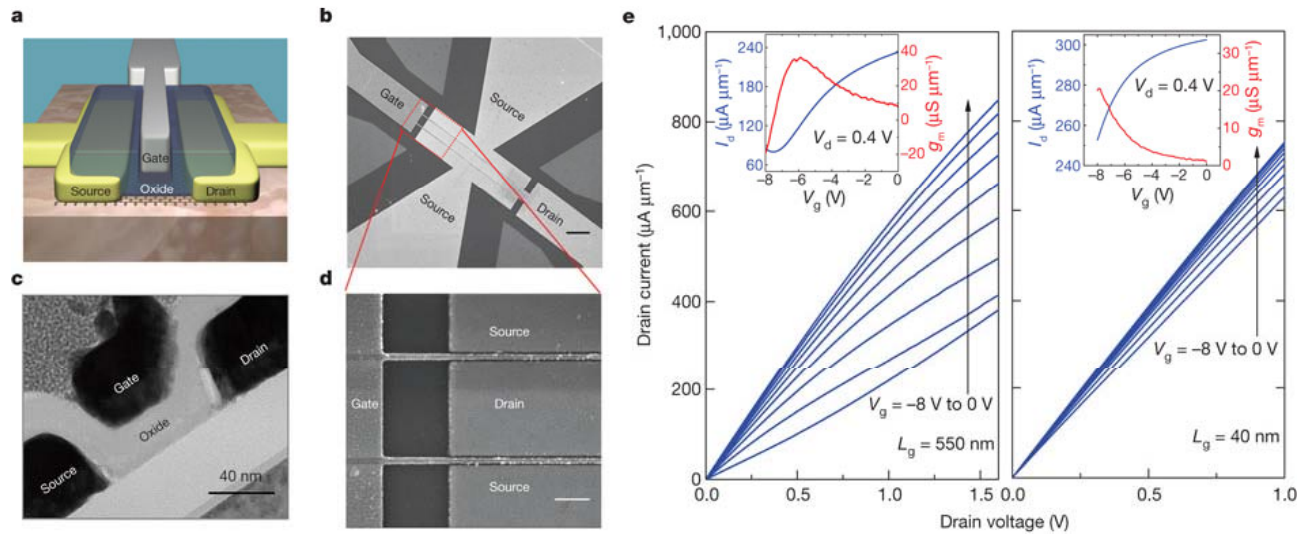
*Proc. Natl. Acad. Sci. U.S.A.* 109, 11588–11592 (2012)

up to **427** GHz

Why use graphene in RF-tech

- Low on-off ratio => fail in CMOS tech
- High mobility, carrier saturation velocity, and large current density=> high cut-off frequency
- High quality graphene => Toward Terahertz regime in the future

# High-frequency, scaled graphene transistors on diamond-like carbon

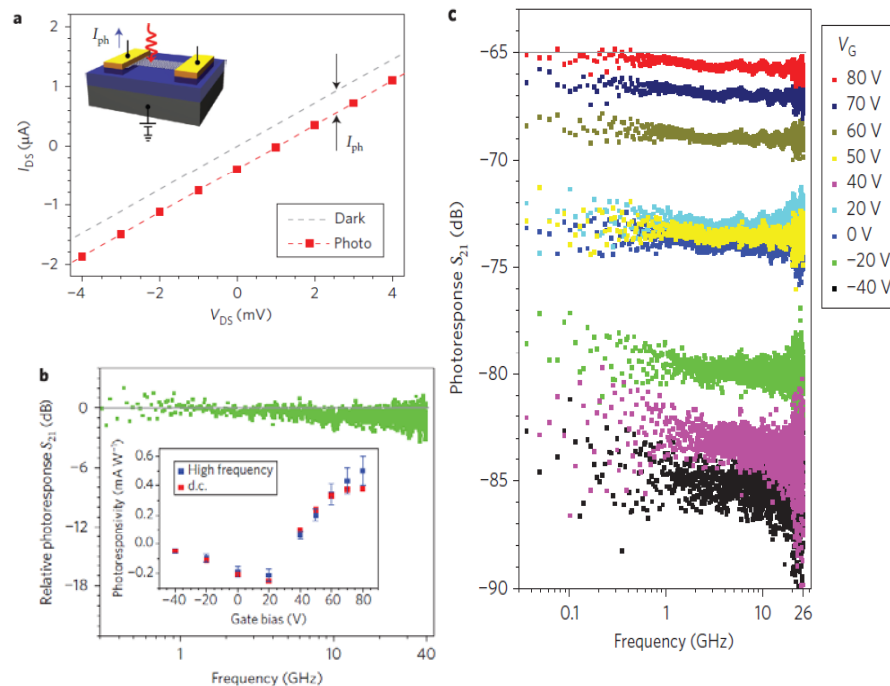


Cut-off frequencies as high as 155 GHz have been obtained for the 40-nm transistors, and the cut-off frequency was found to scale as  $1/(\text{gate length})$ .

Wu, Y., et. al. *Nature*, 2011, 472 74–78

# Ultra fast and Ultra high gain Photodetector

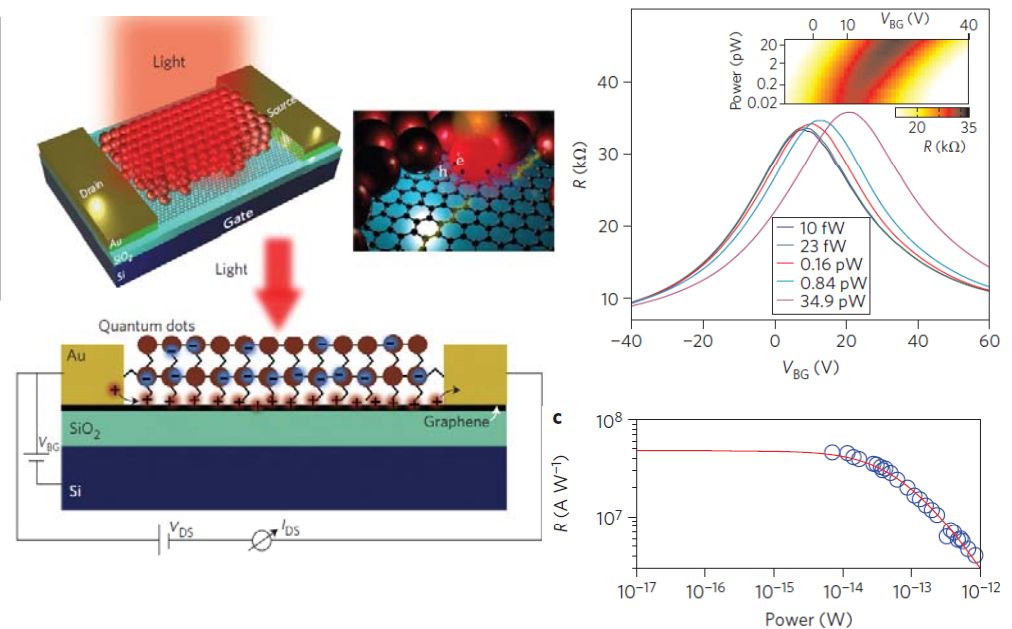
## Ultra fast photodetector



*Nature Nanotech.* 4, 839–843 (2009)

- Light is absorbed by graphene
- P-n junction between channel and graphene covered with metal
- Gate dependent photoresponse
- Could be operated at frequency as high as 40GHz

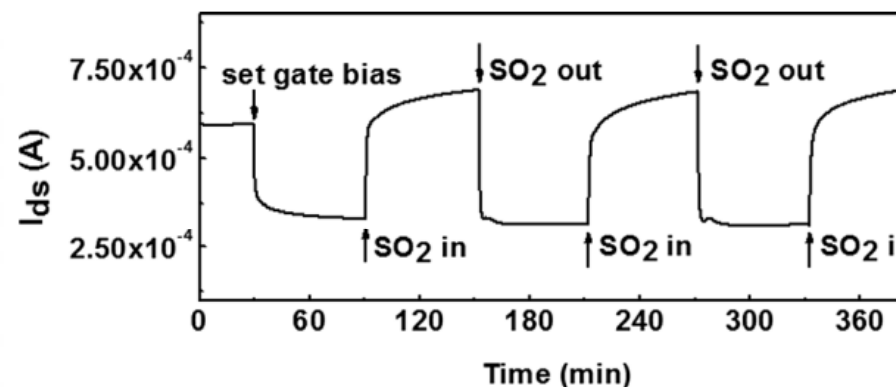
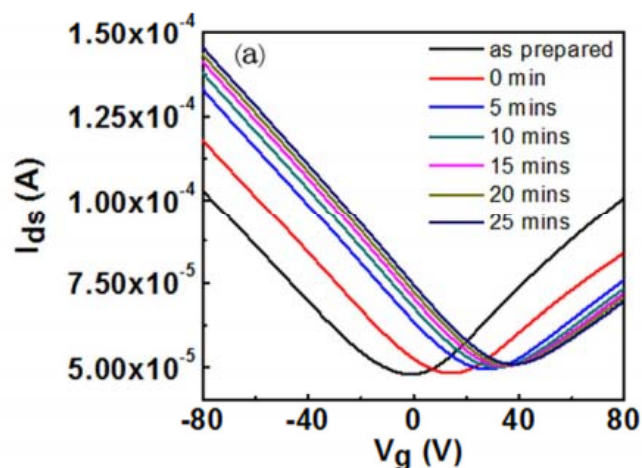
## Ultra high gain photodetector



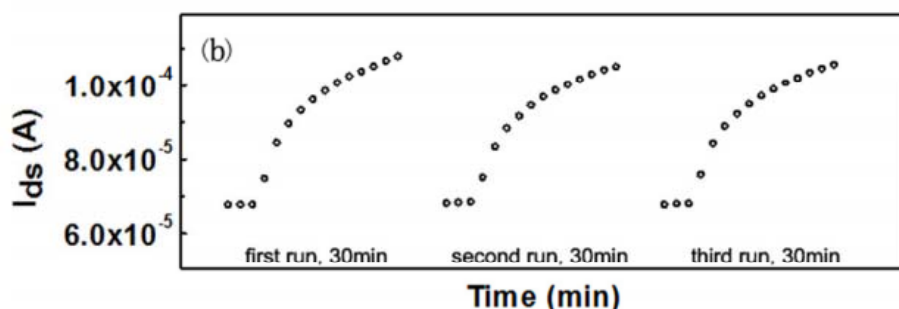
*Nature Nanotech.* 7, 363–368 (2012)

- Light is absorbed by photo-active material on graphene
- Charge transfer from photo-active material to graphene
- Gain could be very large due to inherent high mobility of graphene

# Detection of sulfur dioxide (SO<sub>2</sub>) gas with graphene field effect transistor



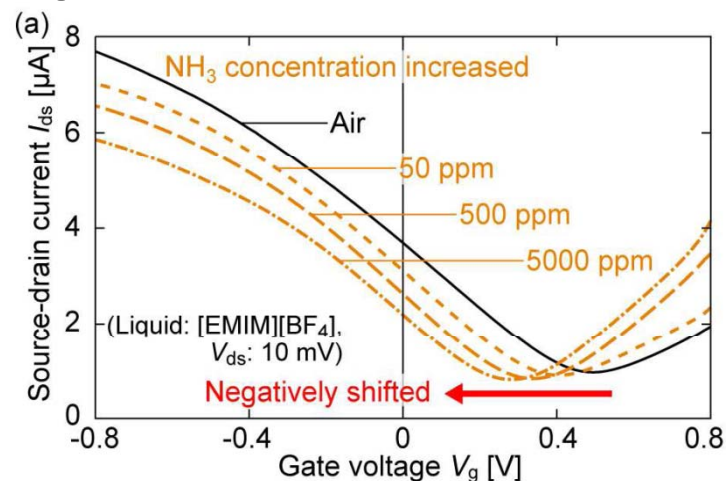
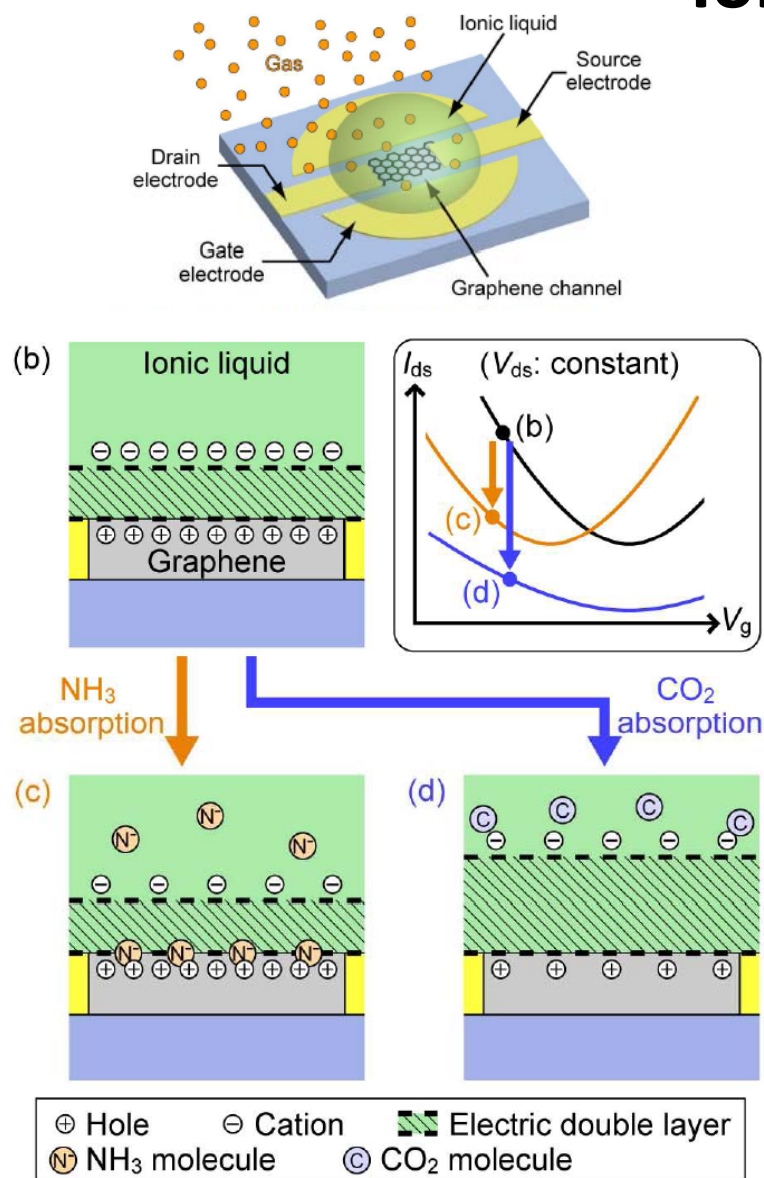
SO<sub>2</sub> strongly p-dopes the graphene



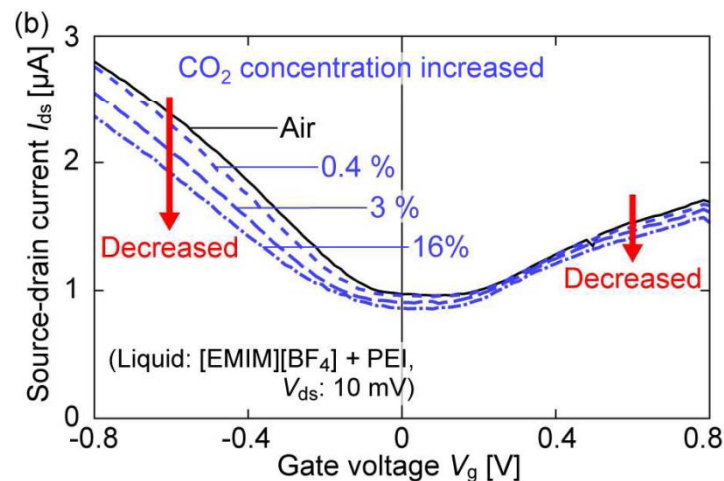
1. the Dirac point shifted 0.678V/ppm to the positive side
2. the resistance decreased about 60% at a SO<sub>2</sub> concentration of 50 ppm.

Reproducibility of the graphene FET as a SO<sub>2</sub> gas detector

# A Graphene FET Gas Sensor ( $\text{CO}_2$ , $\text{NH}_3$ ) Gated by Ionic Liquid

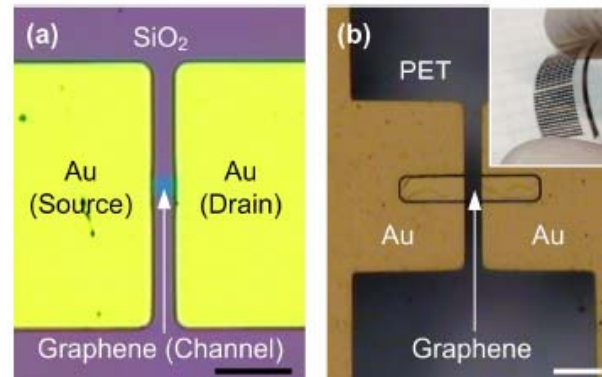
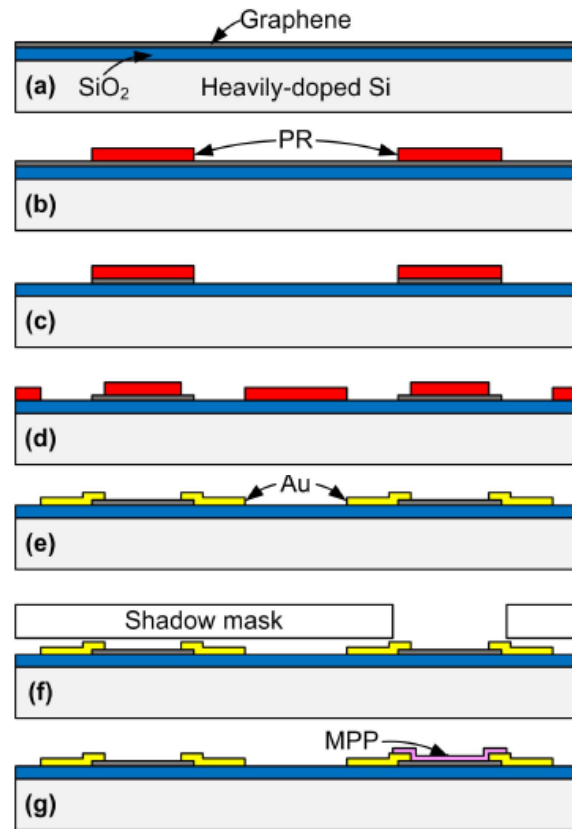


NH<sub>3</sub> molecules transferred negative charge to graphene channel

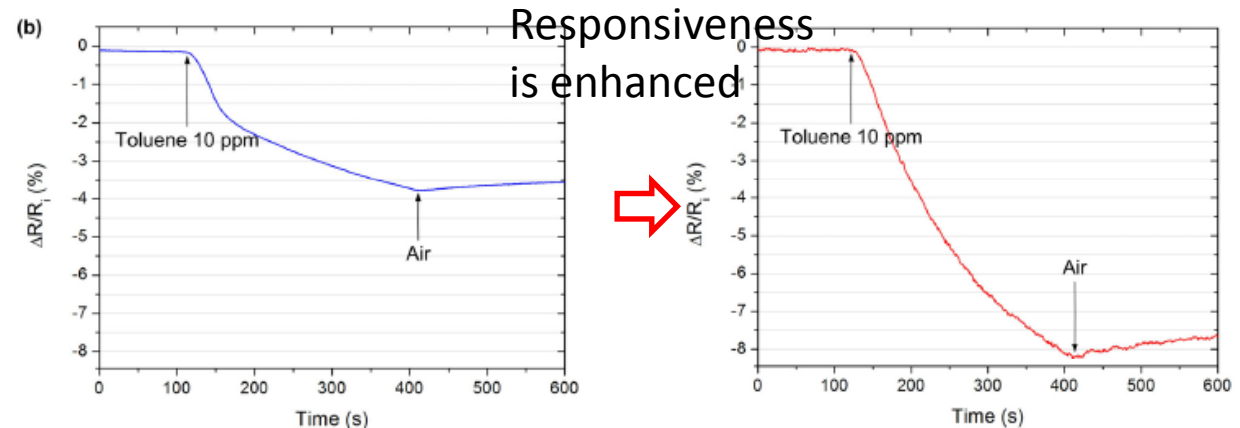


The decrease in  $I_{ds}$  over the entire voltage caused by CO<sub>2</sub> denotes that capacitance of the EDL was reduced.

# Transparent and Flexible Toluene Graphene Sensor



The functionalization of MPP enhanced responsiveness ( $\Delta R/R_i$ ) over 200% compared to pristine graphene



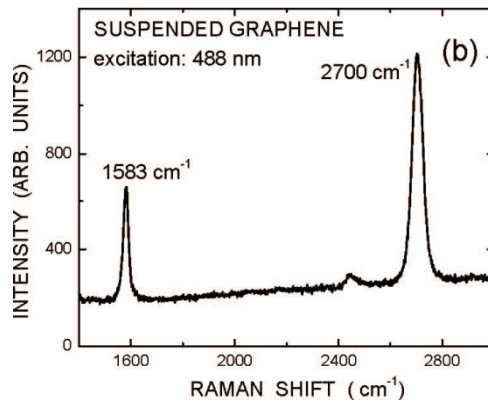
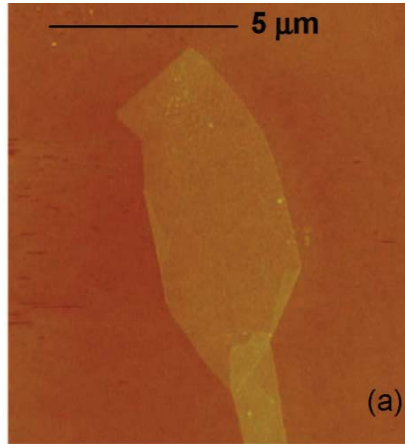
Cobalt-metalloporphyrin (Co-MPP)-functionalized graphene toluene sensor

1. highly sensitive
2. Transparent
3. flexible

Pristine graphene

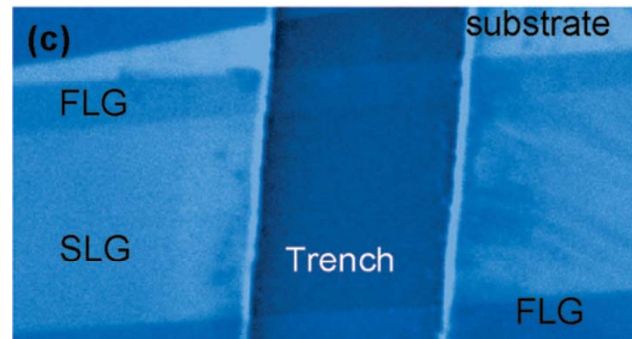
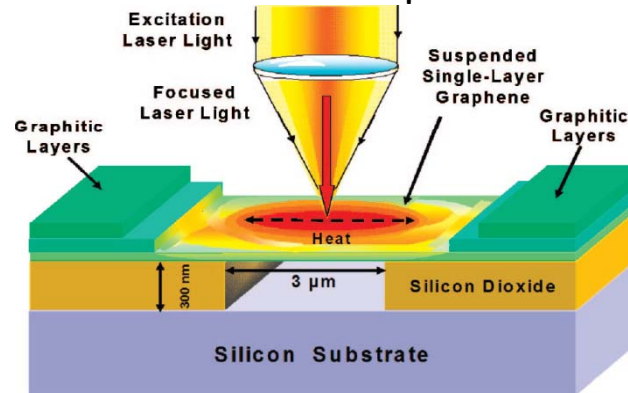
Co-MPP graphene

# Superior Thermal Conductivity of Single-Layer Graphene

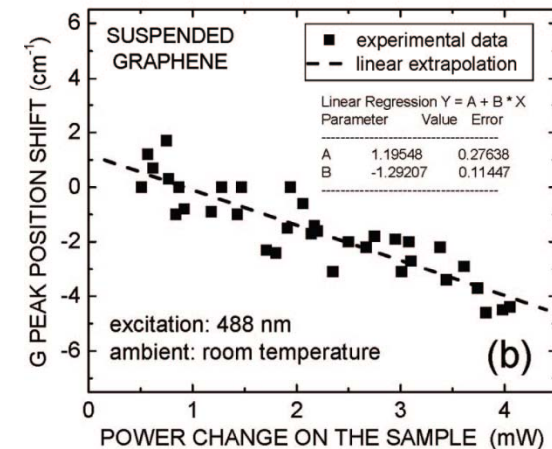
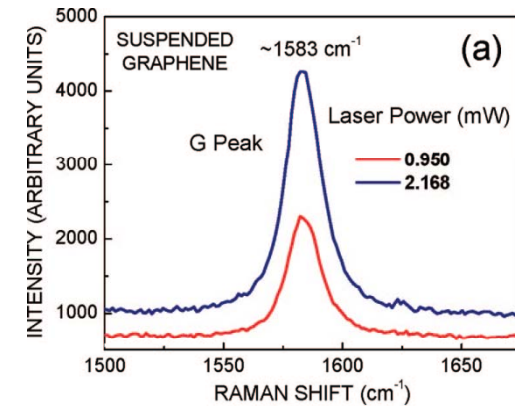


Thermal conductivity of Gold ~ 318 W/mK

Measurement setup

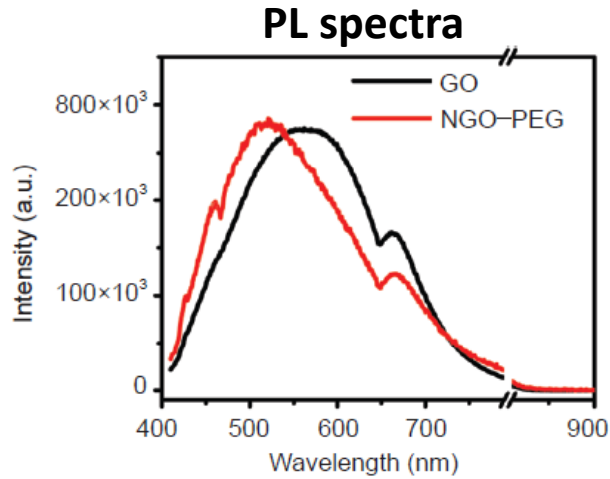
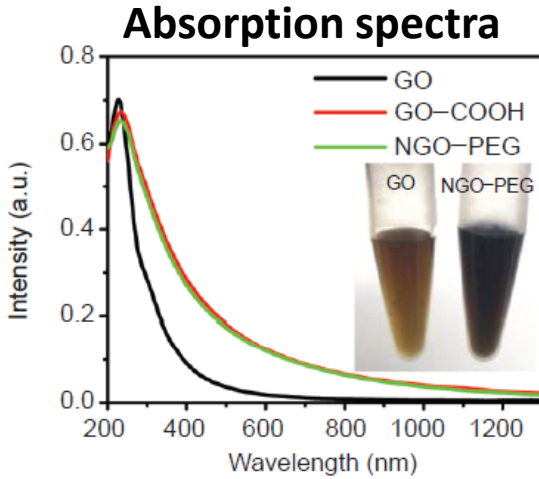


Real device

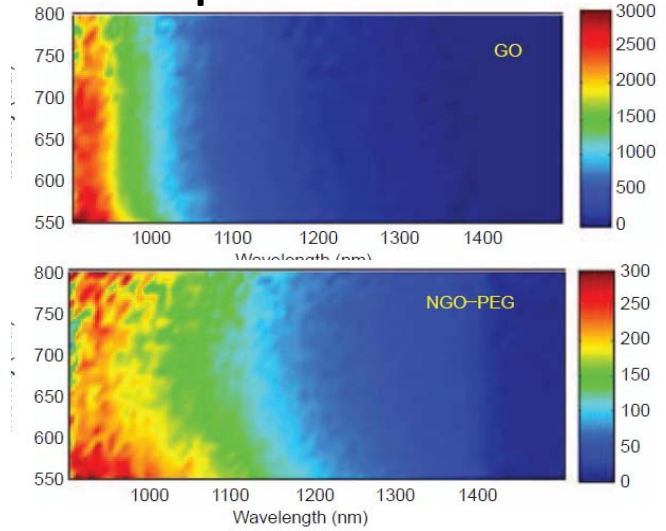


The room temperature values of the thermal conductivity in the range  $\sim (4.84 \times 10^3 \text{ to } 5.30 \times 10^3 \text{ W/mK})$  were extracted from the dependence of the Raman G peak frequency on the excitation laser power and independently measured G peak temperature coefficient

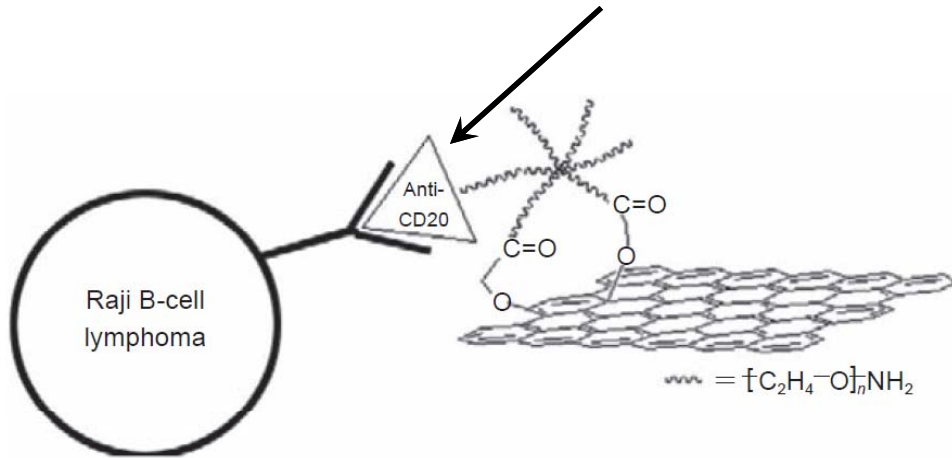
# Functionalized graphene oxide for Cellular Near IR Imaging



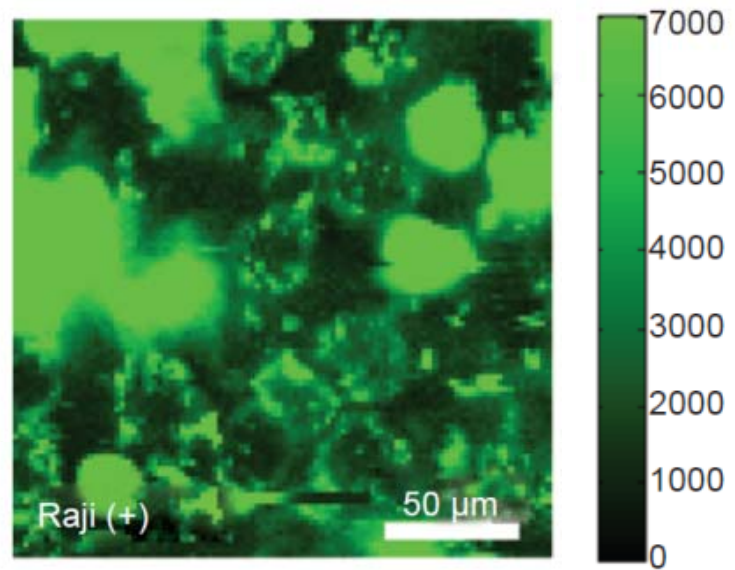
## The PLE spectra of GO and NGO



## Selective binding and cellular imaging of NGO-PEG conjugated with anti-CD20 antibody

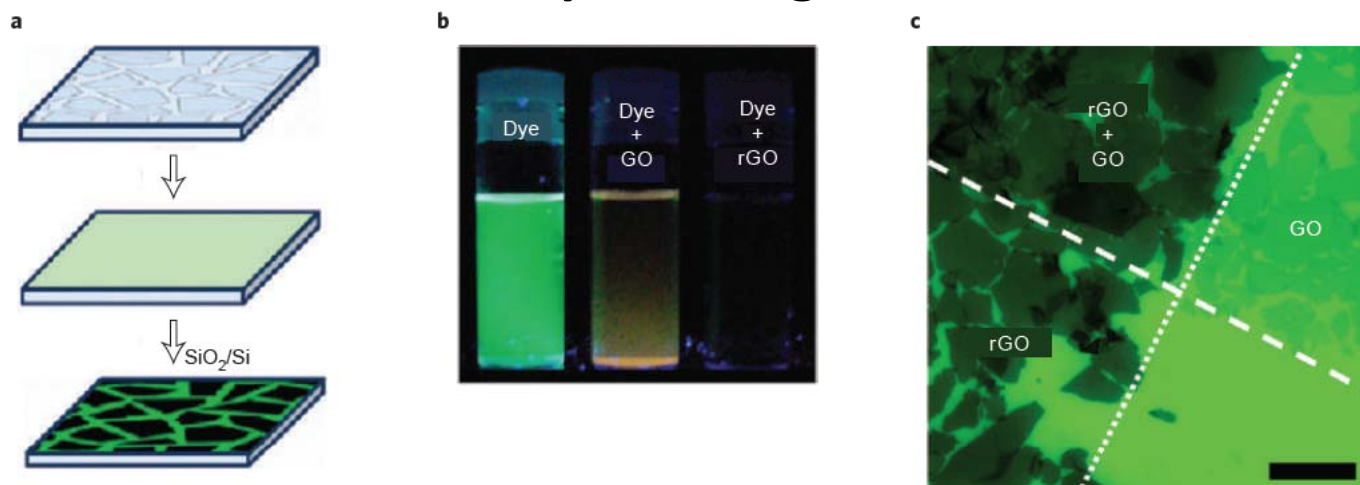


**NIR fluorescence image (1100~2200 nm)**



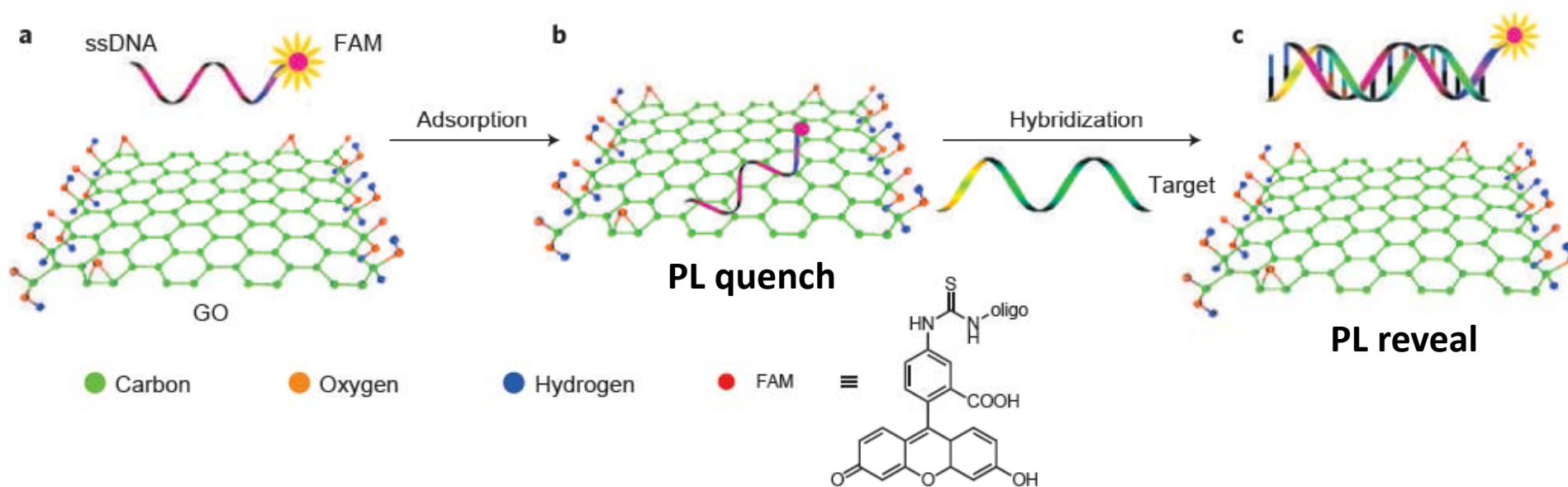
## B-cells

## Fluorescence-quenching of GO and rGO



Kim, J., et. Al., *J. Am. Chem. Soc.* **132**, 260–267 (2010).

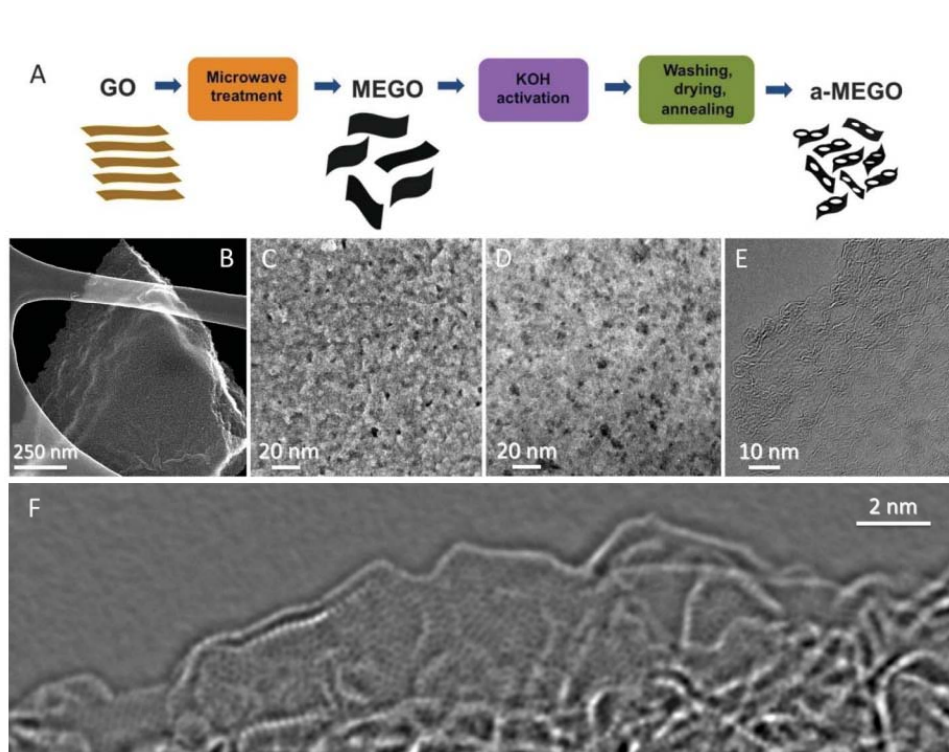
## Biosensing by fluorescence quenching in GO



Lu, C.-H. et. Al., *Angew. Chem. Int. Ed.* **48**, 4785–4787 (2009).

### **3. Energy applications**

# Carbon-Based Supercapacitors Produced by Activation of Graphene

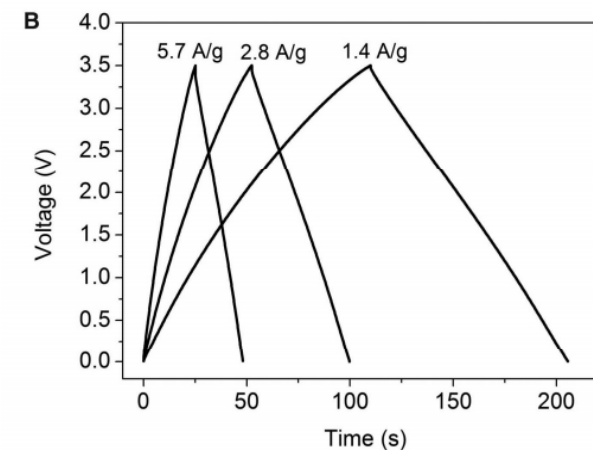
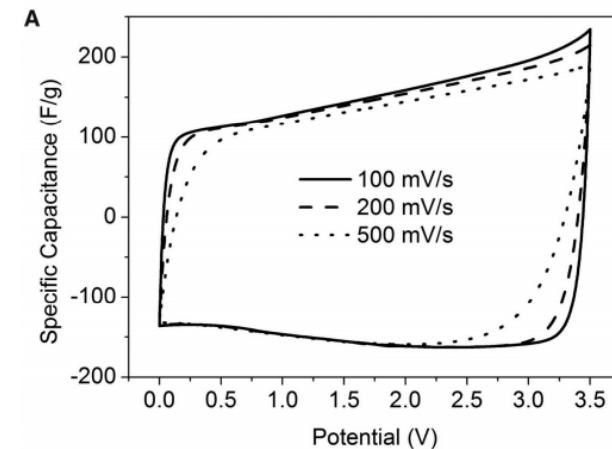


The advantages of MEGO:

High electrical conductivity

Low oxygen and hydrogen content.

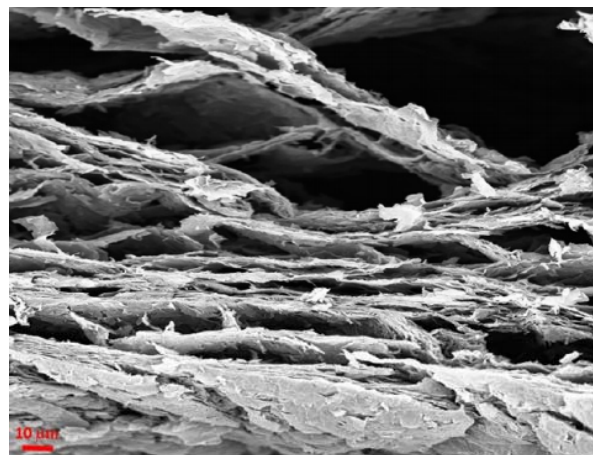
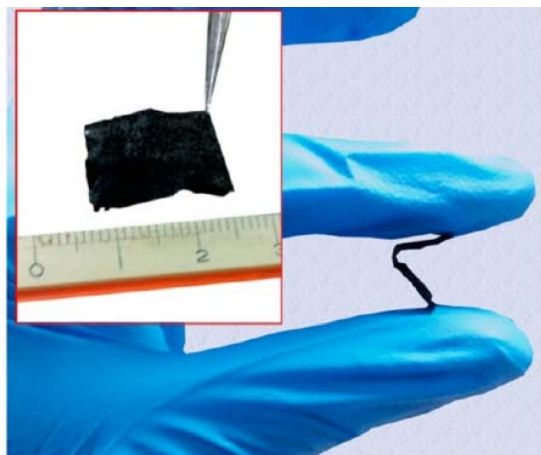
$sp^2$ -bonded carbon with continuous 3D network



For a packaged cell, the power density of **~75 kW/kg** is one order higher than the values from commercial carbon supercapacitors.

# Photothermally Reduced Graphene as High-Power Anodes for Lithium-Ion Batteries

Graphene oxide  $\xrightarrow{\text{Flash light}}$  Reduced graphene oxide

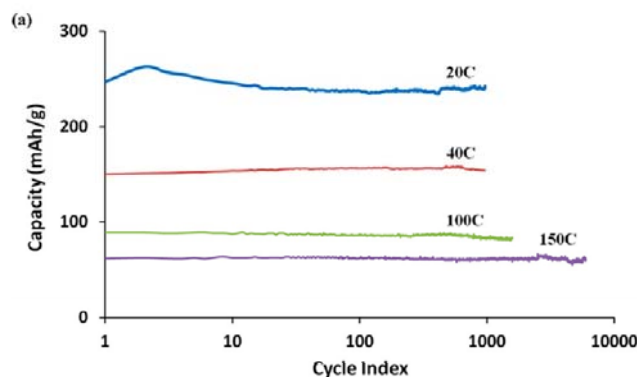
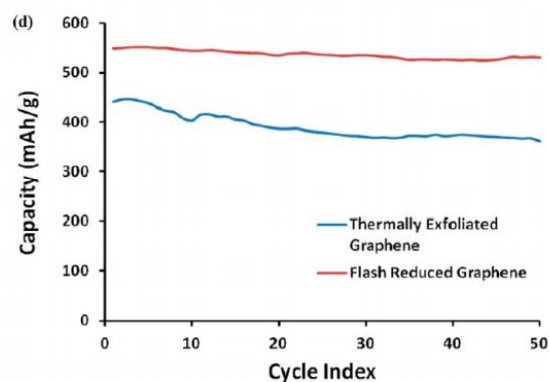


For reduced graphene anodes

At high charge/discharge rates of  $\sim 40$  C,

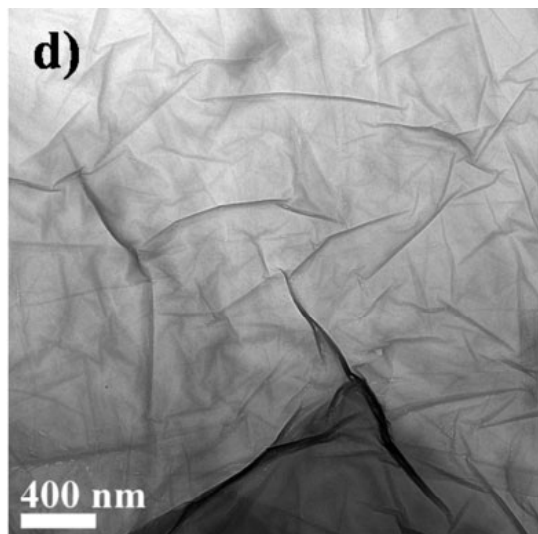
1. A steady capacity of  $\sim 156$  mAh/g over 1000 cycles

2. a stable power density of  $\sim 10$  kW/kg

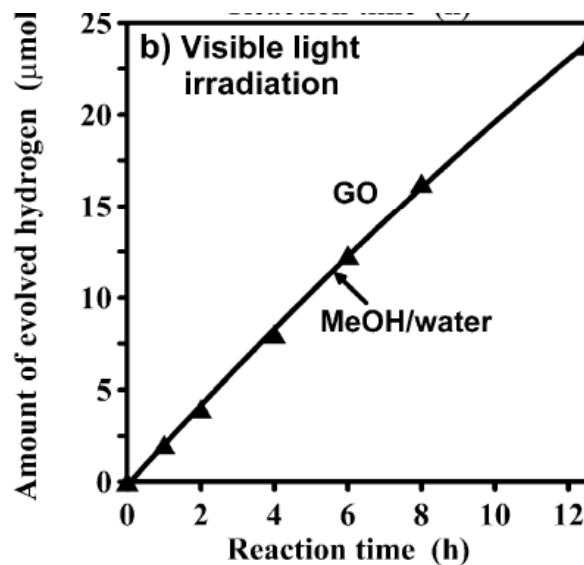
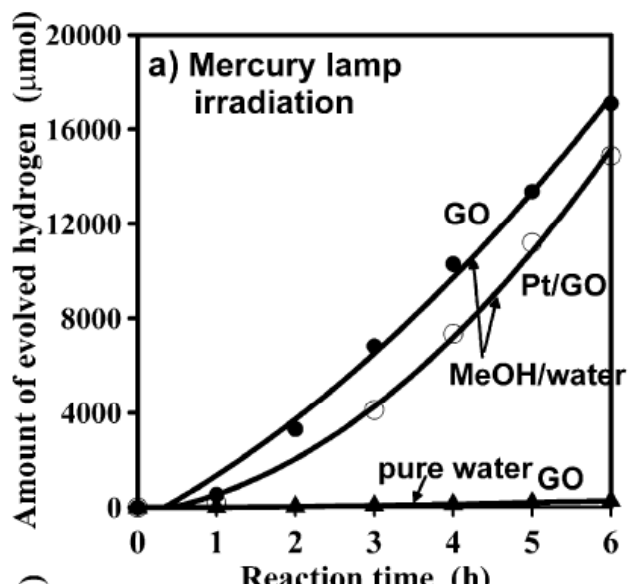
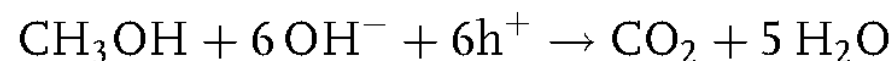
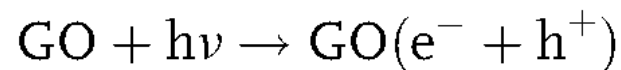
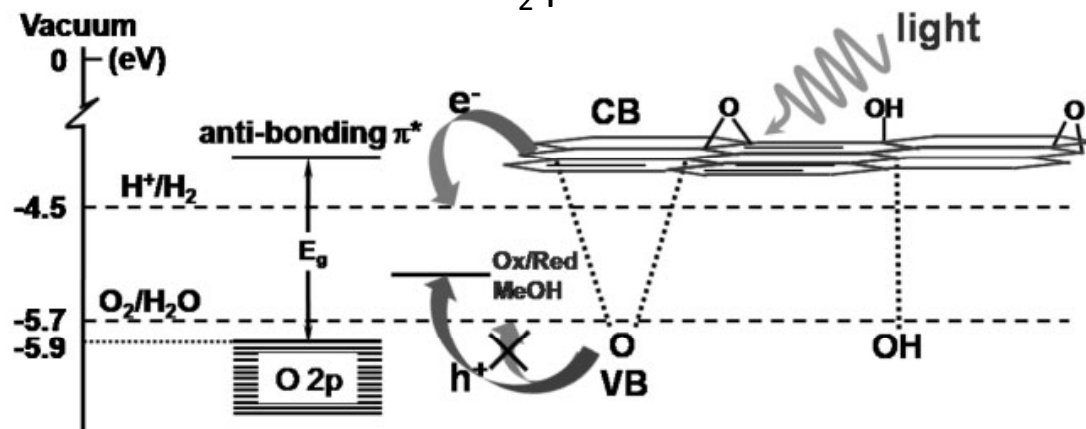


# Graphite Oxide as a Photocatalyst for Hydrogen Production from Water

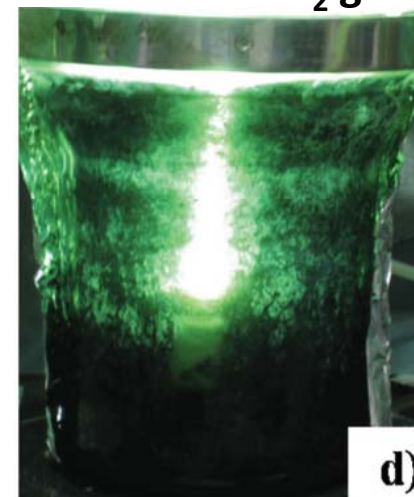
GO sheets



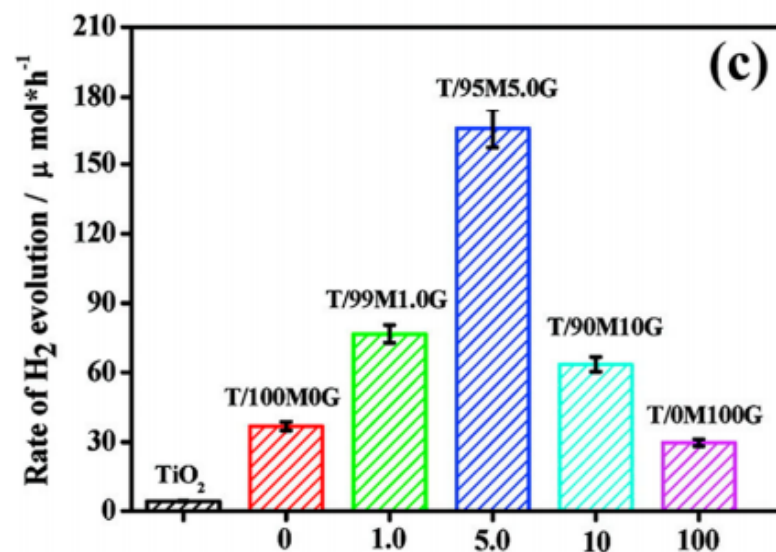
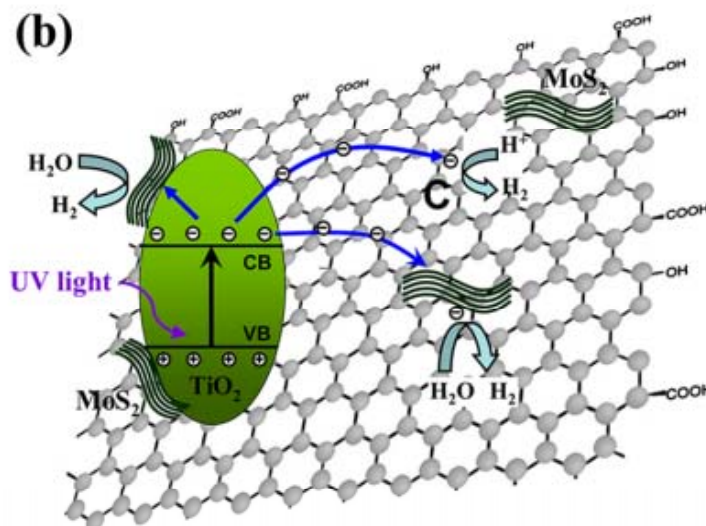
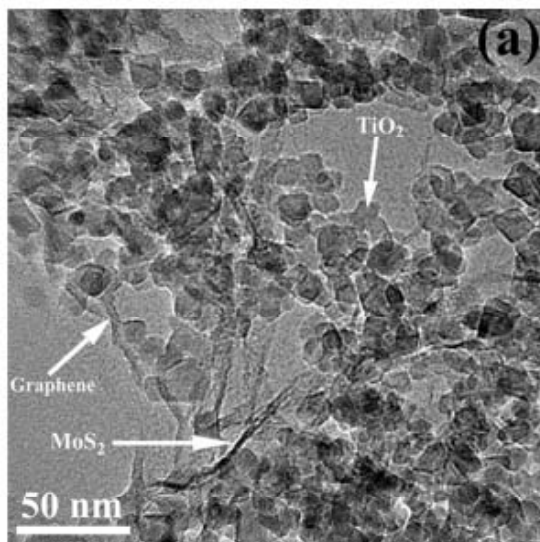
Mechanism of H<sub>2</sub> production from GO sheet



H<sub>2</sub> generation



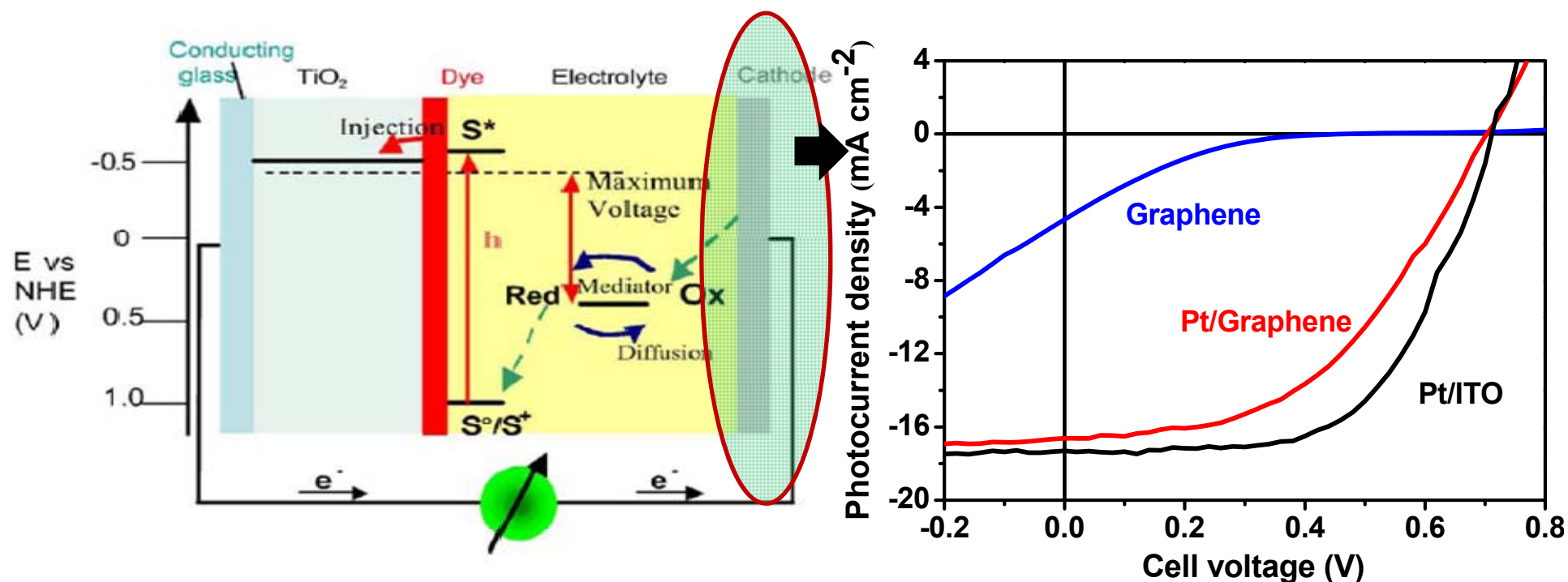
# MoS<sub>2</sub> and Graphene as Cocatalysts for Enhanced Photocatalytic H<sub>2</sub>



The photoexcited electrons are transferred from the CB of TiO<sub>2</sub> not only to the MoS<sub>2</sub> nanosheets but also to the C atoms in the graphene sheets, which can effectively reduce H<sup>+</sup> to produce H<sub>2</sub>.

The TiO<sub>2</sub>/MoS<sub>2</sub>/graphene composite reaches a high H<sub>2</sub> production rate of 165.3  $\mu\text{mol h}^{-1}$ .

# Graphene based counter electrode for DSSCs



| Counter electrode | $V_{oc}$ (V) | $J_{sc}$ (mA) | $FF$ | Efficiency(%) |
|-------------------|--------------|---------------|------|---------------|
| Graphene          | 0.46         | 4.83          | 0.16 | 0.35          |
| Pt/Graphene       | 0.70         | 16.58         | 0.48 | 5.57          |
| Pt/ITO            | 0.71         | 17.40         | 0.59 | 7.29          |

## **4. For environmental and other applications**

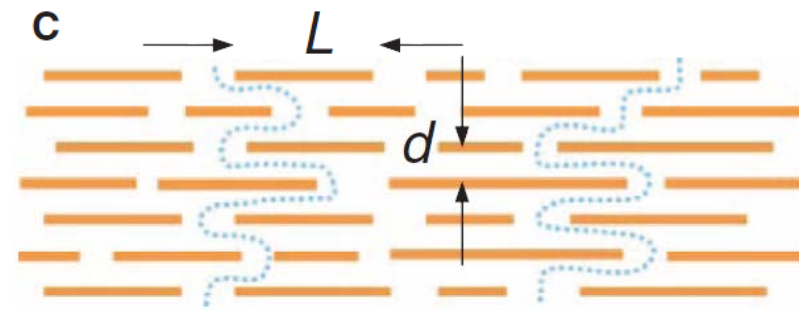
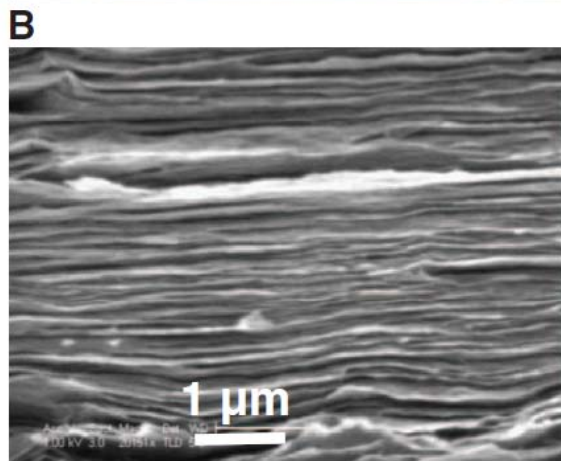
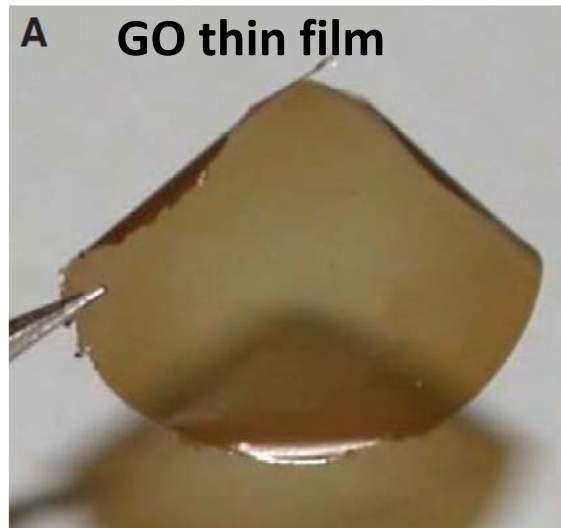
**Permeation of Water**

**→ Desalination of sea water**

**Heavy metal remover**

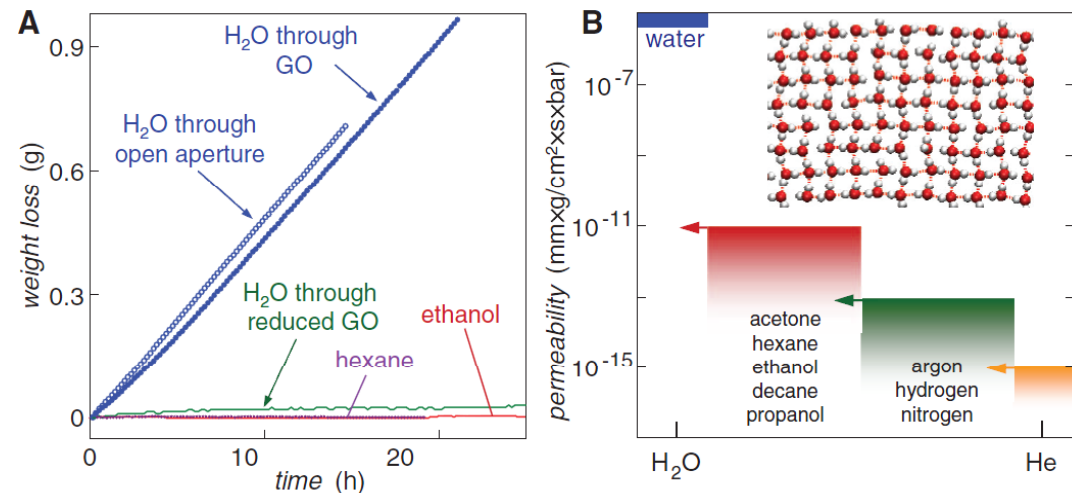
**Superior thermal conductor**

# Unimpeded Permeation of Water Through Helium-Leak-Tight Graphene-Based Membranes



Completely impermeable: liquids, vapors, and gases

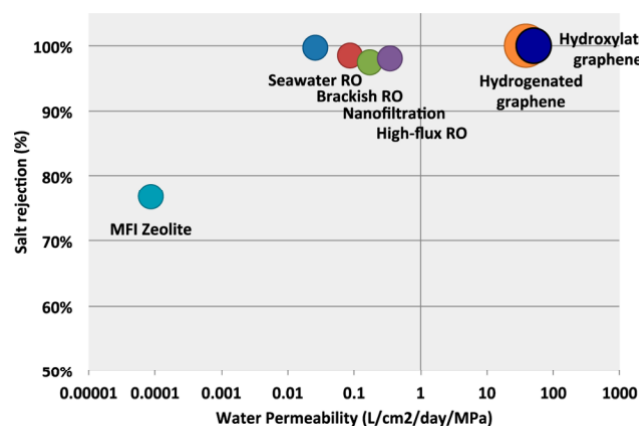
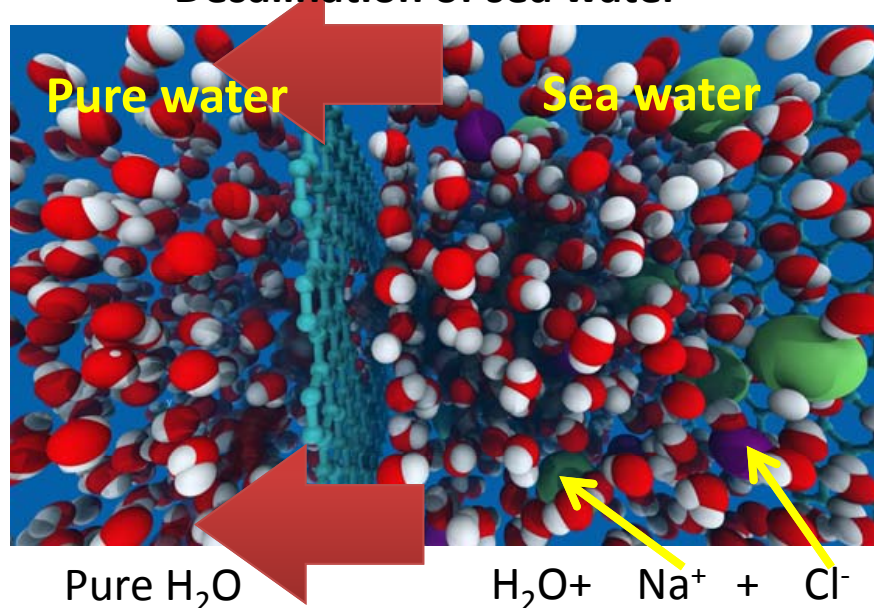
Unimpeded permeation: water H<sub>2</sub>O



H<sub>2</sub>O permeates through the GO at  
least **10<sup>10</sup> times** faster than He gas

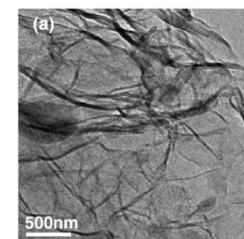
# Graphene sheets for desalination of sea water and arsenic removal

## Desalination of sea water

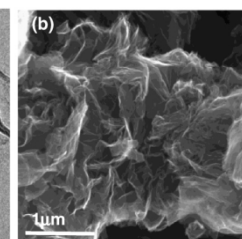


The graphene nanopores reject salt ions with a water permeability 2–3 orders of magnitude higher than commercial RO.

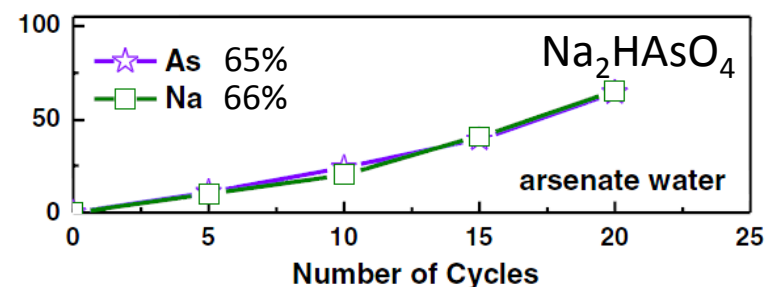
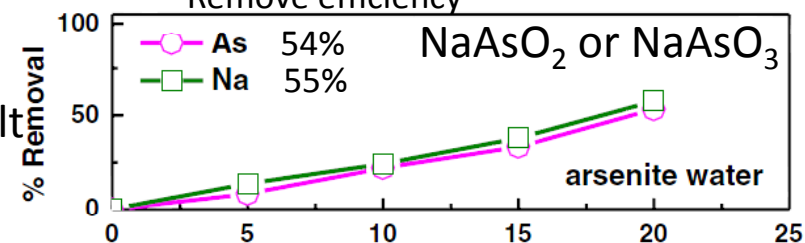
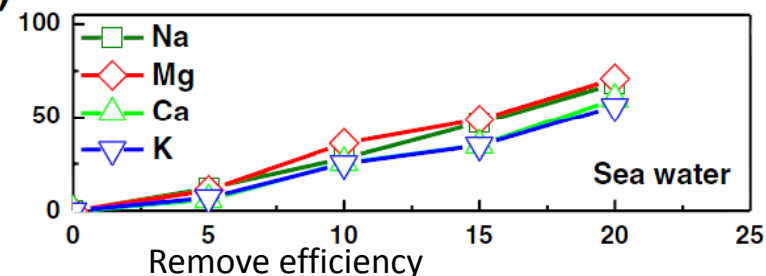
TEM



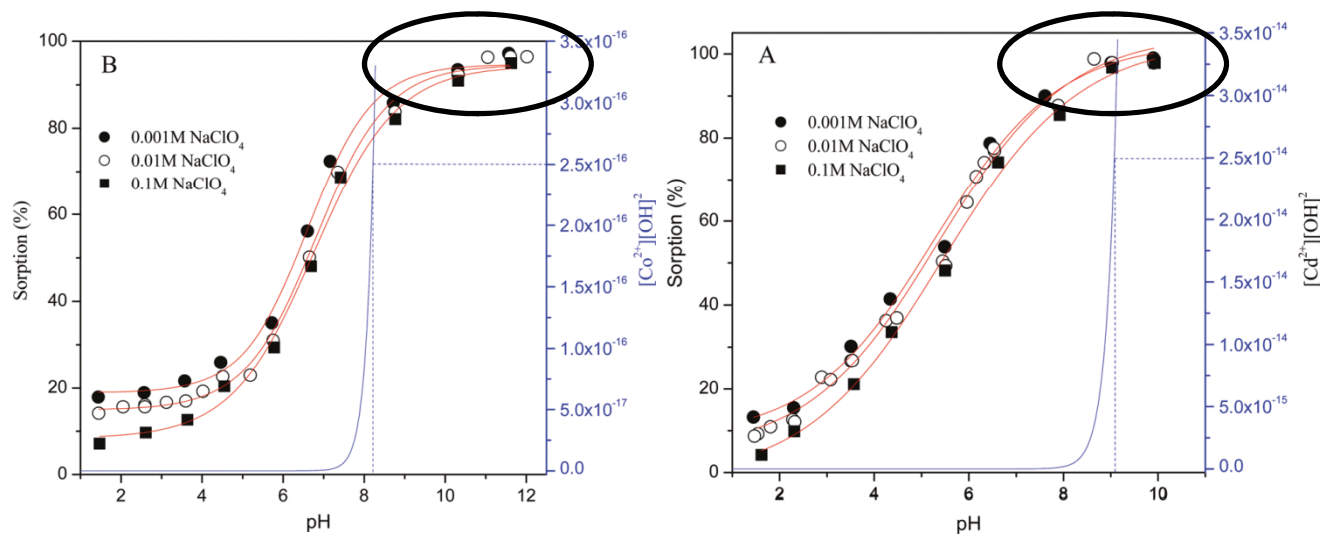
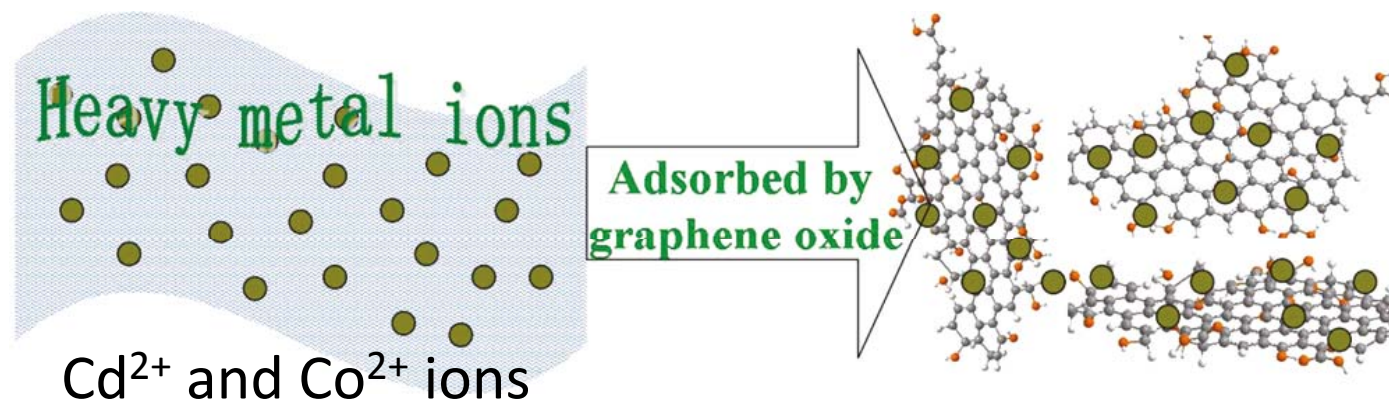
SEM



(c) Removal efficiency for different water



# Few-Layered Graphene Oxide Nanosheets As Superior Sorbents for Heavy Metal Ion Pollution Management



Most  $\text{Cd}^{2+}$  and  $\text{Co}^{2+}$  is adsorbed on GO nanosheets at  $\text{pH} > 9$ .